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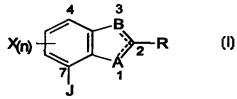
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(54) Title: CYCLOIMIDO-SUBSTITUTED BENZOFUSED HETEROCYCLIC HERBICIDES

(57) Abstract

herbicidal compounds, compositions contain-Novel and methods for their use in controlling ing them. The herbicidal compounds novel weeds disclosed. are represented by formula (I), where is a 1-substituted-6-trifluoromethyl-2,4-pyrimidinedione-3-yl, 1-substituted-6-trifluoromethyl-1,3,5-triazine-2,4-dion-1-yl, a



3,4,5,6-tetrahydrophthalimid-1-yl, a 4-difluoromethyl-4,5-dihydro-3-methyl-1,2,4-triazol-5(1H)-on-1-yl, a 5,6,7,8-tetrahydro-1H,3H-[1,3,4]thiadiazolo[3,5-a]pyridazineimin-1-yl, or a 1,6,8-triazabicyclo[4.3.0]-nonane-7,9-dion-8-yl ring attached at the 7 position of a benzofuran, benzoxazole, indole, 2,3-dihydrobenzimidazole or benzimidazole, and X is selected from hydrogen, halogen, cyano, nitro, and amino. Preferred R groups are optionally substituted alkyl groups.

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WO 98/38188 PCT/US98/03647

CYCLOIMIDO-SUBSTITUTED BENZOFUSED HETEROCYCLIC HERBICIDES

BACKGROUND OF THE INVENTION

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The present invention relates generally to novel herbicidal compounds and methods for their use in controlling unwanted plant species in agriculture. In particular, the present invention pertains to cycloimido-substituted benzofused heterocyclic herbicides, and more particularly it pertains to herbicides in which the 2.3is benzofuran, benzimidazole, benzofused heterocycle а dihydrobenzimidazole, or indole having a cycloimido moiety which is a 1substituted-6-trifluoromethyl-2,4-pyrimidinedione-3-yl, 1-substituted-6а trifluoromethyl-1,3,5-triazine-2,4-dion-1-yl, a 3,4,5,6-tetrahydrophthalimid-1-yl, a 4-difluoromethyl-4,5-dihydro-3-methyl-1,2,4-triazol-5(1H)-on-1-yl, 5.6.7.8tetrahydro-1H,3H-[1,3,4]thiadiazolo[3,5-a]pyridazineimin-1-yl, а 1,6,8triazabicyclo[4.3.0]-nonane-7,9-dion-8-yl ring.

SUMMARY OF THE INVENTION

It has now been found that certain cycloimido-substituted benzofused heterocyclic compounds are useful as pre-emergent and postemergent herbicides.

These novel compounds are represented by formula 1:

where J is a 1-substituted-6-trifluoromethyl-2,4-pyrimidinedione-3-yl, a 1-substituted-6-trifluoromethyl-1,3,5-triazine-2,4-dion-1-yl, a 3,4,5,6-tetrahydrophthalimid-1-yl, a 4-difluoromethyl-4,5-dihydro-3-methyl-1,2,4-triazol-5(1H)-on-1-yl, a 5,6,7,8-tetrahydro-1H,3H-[1,3,4]thiadiazolo[3,5-a]pyridazineimin-1-yl, or a 1,6,8-triazabicyclo[4.3.0]-nonane-7,9-dion-8-yl ring attached at the 7 position of a benzofuran, benzoxazole, 2,3-dihydrobenzimidazole, indole or benzimidazole, and X is selected from hydrogen, halogen, cyano, nitro, alkyl, haloalkyl, and amino. Preferred R groups are optionally substituted alkyl groups.

DETAILED DESCRIPTION OF THE INVENTION

Certain cycloimido-substituted benzofused heterocyclic compounds have now been found to be useful as pre- and postemergent herbicides. These compounds are represented by formula I:

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where

- (1) A is nitrogen double-bonded to position 2 and B is oxygen:
- (2) A is oxygen and B is CR1 double bonded to position 2;
- (3) A is NH and B is nitrogen double-bonded to position 2;
- (4) A is nitrogen double bonded to position 2 and B is NR²;

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- (5) A is CH double bonded to position 2 and B is NR2;
- (6) A is NH and B is CR1 double bonded to position 2; or
- (7) A and B are NH

R is hydrogen, hydroxy, mercapto, straight or branched chain lower alkyl, cycloalkyl, alkoxy, aryl, heteroaryl, alkenyl, haloalkyl, hydroxyalkyl, haloaryl, alkoxyaryl, arylalkyl, aryloxyalkyl, haloarylalkyl, alkylthio, heterocyclyl, alkoxyalkyl, arvicarbonyloxyalkyl, aikyicarbonyloxyaikyl, alkoxylalkyloxyalkyl, aminoalkenyl, carboxy, aminoalkyl, cyanoalkyl, aminocarbonyloxyalkyl, alkylcarboxyalkyl, formyl, aminocarbonyl, amino, carboxyalkyl, alkylcarboxy, aminosulfonyl, alkylsulfonylamino, alkyisulfonyl, nitro, oxygen, cvano, alkoxycarbonylamino, alkylcarboxylalkoxy, alkoxycarbonyloxyalkyl, (aryl)(alkoxy)alkyl, aryliminoalkyl, alkoxycarbonylalkylaminoalkyl, alkynylalkylthio, arylaikoxyaikyl, cyanoalkylthio, (aryl)(alkylcarbonyloxy)alkyl, alkoxycarbonylalkylthio, cyanothioalkyl, cyanothio, arylalkylthio, haloalkylalkynylalkylthio, alkenylalkylthio, aminocarbonylalkylthio, 15 (hydroxy)(aryl)alkyl, arylalkylcarbonylaminoalkyl, aminocarbonyloxyalkyl, aminocarbonylalkyl, alkylsulfonylaminoalkyl, alkylcarbonylaminoalkyl, alkoxycarbonyl, and alkenyloxy, where the amino group may be substituted with one or two substituents independently selected from alkyl, hydroxy, alkoxy, carboxy, aryl, alkylsufonyl, or haloalkylsulfonyl; 20

R¹ is hydrogen, lower alkyl, or haloalkyl;

R² is hydrogen, alkyl, haloalkyl, CO₂(alkyl), CH₂CO₂(alkyl), CH₂CONHalkyl, CH₂CON(alkyl)₂, CH₂CO₂H, CH₂OCH₃, SO₂(alkyl), CH₂CH=CH₂, CH₂C≡CH. X is selected from hydrogen, F, Cl, Br, alkyl, haloalkyl, CN, NO₂, and

NH₂; 25

n is 0-3;

- 4 -

J is selected from

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R³ is selected from hydrogen, alkyl, haloalkyl, CH₂CN, CH₂CH=CH₂, CH₂C≡CH, CH₂CO₂(alkyl), CH₂OCH₃, and NH₂.

Preferred compounds are those of formula I where R is CH_3 , CH_2CH_3 , $C(CH_3)_2OH$, CH_2CH_2OH , $CH(CH_3)_2$, t-butyl, CF_3 , $CH(F)CH_3$, CF_2CF_3 , $C(CH_3)_2OCOCH_3$, $C(CH_3)_2HSOCH_3$, $CHCH_2CHCH_2CHCH_3$ CHCHCQCH, 2 and $CON(CH_3)_2$; X is a chlorine, bromine or fluorine substituted in one or both of positions 4 and 6; J is

$$O \bigvee_{N-R^3}^{N} O \\ CF_3$$

and R³ is CH₃ or NH₂.

One aspect of the present invention relates to compounds of formula I in which A is nitrogen double-bonded to position 2 and B is oxygen, and R, R³, J, X and n are as described above.

Another aspect of the present invention relates to compounds of formula I in which A is oxygen and B is CR^1 double bonded to position 2, and R, R^1 , R^3 , J, X and n are as described above.

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Another aspect of the present invention relates to compounds of formula I in which A is NH and B is nitrogen double-bonded to position 2, and R, J, X and n are as described above.

Another aspect of the present invention relates to compounds of formula I in which A is nitrogen double bonded to position 2 and B is NR², and R, R², R³, J, X and n are as described above.

Another aspect of the present invention relates to compounds of formula I in which A is CH double bonded to position 2 and B is NR^2 , and R, R^2 , R^3 , J, X and n are as described above.

Another aspect of the present invention relates to compounds of formula I in which A is NH and B is CR¹ double bonded to position 2, and R, R¹, R³, J, X and n are as described above.

Another aspect of the present invention relates to compounds of formula I in which A and B are NH and R, R¹, R³, J, X and n are as described above.

Another aspect of the present invention relates to compounds of formula I where J is not

$$O \bigvee_{N-R^3} O$$

$$CF_3$$

when: A is oxygen and B is CR¹ double bonded to position 2; A is CH double bonded to position 2 and B is NR²; or A is NH and B is CR¹ double bonded to position 2; and R, R¹, R³, X, and n are as described above.

As shown in the specification a wide range of substituents is described for position B in compounds of formula I whereas position A is generally unsubstituted. It was found that some herbicidal activity is retained when a methyl substituent is placed at position A, but that substitution at that position generally causes a sharp decrease in activity.

Certain intermediates of the present invention are novel. These include compounds of formula II:

PCT/US98/03647

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where Y is NO_2 , NH_2 or -NHN=C(CH $_3$ R; Z is hydrogen, F, NH $_2$ or OH; and R, J, X, and n are as described above; with the proviso that when Y is - NHN=C(CH $_3$)R, Z is hydrogen.

As used in this specification and unless otherwise indicated, the terms "alkyl," "alkenyl," "alkynyl," "haloalkyl," and "alkoxy" used alone or as part of a larger moiety, includes straight or branched carbon chains of 1 to 6 carbon atoms. "Halogen" refers to fluorine, bromine or chlorine. "THF" means tetrahydrofuran, "DMF" means N,N-dimethylformamide, and "DBU" means 1,8-diazabicyclo[5.4.0]undec-7-ene. When "n" in "X_(n)" is 2 or 3, the substituents X may be the same or different from one another.

- 7 -

Scheme 1

a) 70% HNO₃/H₂SO₄, 0-5 ∘C; (b) NaOSi(CH₃)₃, MeOH, dioxane; (c) Fe, EtOH, acetic acid, HCl, heat; (d) CF₃C(NH₂)=CO₂CH₂CH₅, NaOSi(CH₃)₃, DBU, DMF; (e) CH₃I, K₂CO₃, DMF, 60-80 °C; (f) HCl, NaNO₂ NaI, H Q; (g) BBr ,₃ CH QI ;₂ (h) HC≡CR, Pd(Ph₃P) ₂CI₂, CuI, triethylamine.

bonded to position 2, may be generally prepared as shown in Scheme 1. Starting with an appropriately substituted fluoroaniline derivative 1, nitration provides intermediate 2. Displacement of the fluorine of 2 with a methoxy group as shown in step b, followed by reduction of the nitro group as shown in step c provide the methoxyaniline 3. The methoxyaniline 3 is a versatile intermediate from which a number of compounds of the present invention can be made by attachment of various J groups. For example, a uracil ring may be appended as shown in step d to give intermediate 4a. At this point, R³ substituents other than H may be introduced, as shown for example in step e to provide 4b where R³ is methyl.

Using diazotization conditions (step f) 4b is converted to the iodoanisole 5 which is then deprotected to give the iodophenol 6. Palladium-catalyzed acetylenic coupling and ring closure as shown in step h give benzofurans 7 of the present invention. To obtain benzofurans of formula I where the J group is other than uracil, approaches analogous to that outlined in Scheme 1 may be followed. Such approaches based on Scheme 1 would be known to one skilled in the art.

Scheme 2

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a) 70% HNO₃/H₂SO₄ 0-5 °C; (b) Fe, aqueous acetic acid, 50 °C; (c) RCOCl, pyridinium p-toluenesulfonate, triethylamine, xylene; (d) 1,1-carbonylimidazole, THF; (e) R'-halide, Ag₂O, CH₂Cl₂ (to give **11** where R=R'O).

Benzoxazoles of formula I, where A is nitrogen double bonded to position 2 and B is oxygen, may be prepared as shown in Scheme 2 above. Starting with a phenol such as intermediate 8 nitration under standard conditions

gives the nitrophenol 9. Certain of the benzoxazoles 11 of the present invention may be obtained by reduction of 9 to the aniline 10 followed by treatment with an acid halide (such as shown in step c). Alternatively, other benzoxazoles 11 may be obtained by treating 10 with carbonyldiimidazole to give intermediate 12 which can be O-alkyated according to step e. The approach outlined in Scheme 2 can be adapted, in ways known to one skilled in the art, to obtain benzoxazoles of formula I where the J group is other than uracil.

Scheme 3

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a) see steps (d) and (e) of Scheme 1; (b) 70% HNO_3/H_2SO_4 , 0-5 °C; (c) NH_4OAc , triethylamine, dioxane, heat; (d) $SnCl_2H_2O$ or Fe, NH_4Cl , aqueous ethanol, heat; (e) RCO_2H , heat; RCO-halide, CH_2Cl_2/Py ridine, then $POCl_3$, CH_2Cl_2 ; alkoxycarbonyl isothiocyanate, $HgCl_2$, heat (where R is -NHCO2alkyl); or thiophosgene, EtOAC, heat (where R is -SH).

Benzimidazoles of formula I, where A is NH and B is nitrogen double bonded to position 2, may be prepared as shown in Scheme 3 above. For example,

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intermediate 13 may be converted to the uracil 14 by the well-known chemistry previously described. Nitration of 14 followed by aminolysis of the fluorine group (steps b and c) provides the nitroaniline 15. The diamine 16 is obtained by reduction of 15 under standard conditions. Benzimidazoles 17 of the present invention are obtained by treatment of 16 with a carboxylic acid, an acid halide, an alkoxycarbonyl isothiocyanate, or thiophosgene according to step e. Other benzimidazoles 17 of the present invention are obtained by derivativization of benzimidazoles depicted in Scheme 3 using techniques known to one skilled in the art. The approach outlined in Scheme 3 can be adapted, in ways known also to one skilled in the art, to obtain benzimidazoles of formula I where the J group is other than uracil.

$$\begin{array}{c|c}
X & O \\
N & N \\
N & N \\
N & CF_3
\end{array}$$

Benzimidazoles of structure 17A where R³ is NH₂ are prepared in a manner analogous to that depicted in Scheme 3, except the NH₂ group is attached following nitration of the phenyl ring. The 1-unsubstituted uracil ring is formed as previously described in step d of Scheme 1, followed by nitration of the phenyl ring (Scheme 3, step b). The uracil ring is then aminated in the 1-position by methods known in the art by treating it with 1-aminooxysulfonyl-2,4,6-trimethylbenzene. The 1-aminouracil is then subjected to aminolysis of the phenyl fluorine (step c) followed by reduction to the diamine (step d).

17B

2,3-Benzimidazoles of formula I, where A and B are NH may be prepared from Intermediate 16 in Scheme 3 by heating it with an appropriately substituted acetaldehyde ethyl hemiacetal, affording compounds of Structure 17B.

Scheme 4

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a) i. NaNO₂, HCl; ii. SnCl₂'2H₂O; iii. RCOCH₃ (b) polyphosphoric acid, 80 °C.

Indoles of formula I, where A is CH double bonded to position 2 and B is NR¹, may be prepared according to Scheme 4 above. Using a Fischer indole route the starting aniline 18 may be converted to the corresponding hydrazone 19 which in turn may be cyclized under acidic conditions such as is shown in step b. The resulting indoles 20 of the present invention may be further derivatized by alkylation of the indole ring nitrogen to indoles of formula I where R¹ is other than hydrogen. The approach outlined in Scheme 4 can be adapted, in ways known to one skilled in the art, to obtain indoles of formula I where the J group is other than uracil.

Indoles of formula I, where A is NH and B is CR¹ double bonded to position 2, may be prepared by a Fischer indole synthesis analogous to that shown in Scheme 4 starting with aniline 21. Substitution at the 3 position of indoles such as 22 with R¹ groups is known to one skilled in the art.

Compounds of the present invention may also be prepared in accordance with the procedures shown in the Examples below, by procedures analogous to those shown in the Examples, or by other methods that are generally known or available to one skilled in the art.

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EXAMPLE 1

1-METHYL-6-TRIFLUOROMETHYL-3-[7-BROMO-5-FLUORO-2-(2-METHYLCARBONYLOXYPROP-2-YL)BENZOXAZOL-4-YL]-2,4(1H,3H)-PYRIMIDINEDIONE (COMPOUND 104)

Step A 1-methyl-6-trifluoromethyl-3-(4-bromo-2-fluoro-5-hydroxy-6-nitrophenyl)-2,4(1H,3H)-pyrimidinedione

A stirred solution of 17.0 grams (0.044 mole) of 1-methyl-6-trifluoromethyl-3-(4-bromo-2-fluoro-5-hydroxyphenyl)-2,4(1H,3H)-pyrimidinedione and 5.0 grams (0.050 mole) of sulfuric acid in 100 mL of glacial acetic acid was cooled to 15 °C, and 3.2 grams (0.050 mole) of 70% nitric acid was added dropwise. The reaction mixture was then allowed to warm to ambient temperature where it stirred for two hours. The reaction mixture was poured into water and extracted with diethyl ether. The extract was concentrated under reduced pressure to a residue. The residue was purified by column chromatography on silica gel, yielding 16.4 grams of title compound; mp 76-78 °C.

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Step B 1-methyl-6-trifluoromethyl-3-(6-amino-4-bromo-2-fluoro-5-hydroxyphenyl)-2,4(1H,3H)-pyrimidinedione

A stirred solution of 16.0 grams (0.037 mole) of 1-methyl-6-trifluoro-methyl-3-(4-bromo-2-fluoro-5-hydroxy-6-nitrophenyl)-2,4(1H,3H)-pyrimidinedione and 10 mL of water in 120 mL of glacial acetic acid was heated to 50 °C, and 16.0 grams (excess) of iron dust was slowly added. The reaction mixture was then cooled to ambient temperature where it stirred for one hour. The reaction mixture was filtered through diatomaceous earth, and the filtrate was partitioned in a mixture of 150 mL portions each of water and ethyl acetate. The organic layer was separated, dried with magnesium sulfate, and filtered. The filtrate was concentrated under reduced pressure to a residue. The residue was purified by column chromatography on silica gel, yielding 12.0 grams of title compound; mp 98-100 °C.

Step C Compound 104

A stirred solution of 0.50 gram (0.0013 mole) of 1-methyl-6-trifluoro-methyl-3-(6-amino-4-bromo-2-fluoro-5-hydroxyphenyl)-2,4(1H,3H)-pyrimidinedione, 0.21 gram (0.0013 mole) of 1-chlorocarbonyl-1-methylethyl acetate, 0.14 gram (0.0014 mole) of triethylamine, and 0.16 gram (0.0006 mole) of pyridinium ptoluenesulfonate in 50 mL of xylene was heated at 150 °C for about 18 hours. The reaction mixture was then cooled to ambient temperature and taken up in ethyl acetate. The solution was washed with water and an aqueous solution saturated with sodium chloride; then it was dried with magnesium sulfate. The mixture was filtered, and the filtrate was concentrated under reduced pressure to a residue. The residue was purified by column chromatography on silica gel, yielding 0.72 gram of Compound 104. The NMR spectrum was consistent with the proposed structure.

EXAMPLE 2

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1-METHYL-6-TRIFLUOROMETHYL-3-(7-BROMO-5-FLUORO-2-METHOXY-BENZOXAZOL-4-YL)-2,4(1H,3H)-PYRIMIDINEDIONE (COMPOUND 109)

Step A 1-methyl-6-trifluoromethyl-3-(7-bromo-5-fluorobenzoxazol-2-on-4-yl)-2,4(1H,3H)-pyrimidinedione

A stirred solution of 2.0 grams (0.005 mole) of 1-methyl-6-trifluoromethyl-3-(6-amino-4-bromo-2-fluoro-5-hydroxyphenyl)-2,4(1H,3H)-pyrimidinedione and 1.2 grams (0.008 mole) of carbonylimidazole in 50 mL of **THF** was heated at reflux for three hours. The reaction mixture was cooled and concentrated under reduced pressure to a residue. The residue was purified by column chromatography on silica gel, yielding 1.1 grams of title compound. The NMR spectrum was consistent with the proposed structure.

Step B Compound 109

A mixture of 0.50 gram (0.001 mole) of 1-methyl-6-trifluoromethyl-3-(7-bromo-5-fluorobenzoxazol-2-on-4-yl)-2,4(1H,3H)-pyrimidinedione 0.17 gram (0.001 mole) of methyl iodode, and 0.27 gram (0.001 mole) of silver(I) oxide in 50 mL of methylene chloride was stirred at ambient temperature for two hours. The product was isolated from the reaction mixture by column chromatography on silica gel, yielding 0.28 gram of Compound 109. The NMR spectrum was consistent with the proposed structure.

EXAMPLE 3

1-METHYL-6-TRIFLUOROMETHYL-3-[7-CHLORO-5-FLUORO-2-(1-METHYLETHYL)BENZOXAZOL-4-YL]-2,4(1H,3H)-PYRIMIDINEDIONE (COMPOUND 28)

25 Step A 1-methyl-6-trifluoromethyl-3-(4-chloro-2-fluoro-5-hydroxyphenyl)-2,4(1H,3H)-pyrimidinedione

A stirred solution of 18.2 grams (0.054 mole) of 1-methyl-6-trifluoromethyl-3-(5-amino-4-chloro-2-fluorophenyl)-2,4(1H,3H)-pyrimidinedione in 100 mL of sulfuric acid was cooled to 5 °C, and a solution of 3.7 grams (0.054 mole) of sodium nitrite in about 10 mL of water was added dropwise. The reaction mixture was then warmed to ambient temperature where it stirred for two hours.

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In a separate reaction vessel, a stirred mixture of 242 grams (0.970 mole) of copper(II) sulfate and 1.5 grams (0.005 mole) of iron(II) sulfate heptahydrate in about 300 mL of water and 300 mL of xylene was heated to reflux, and the pyrimidinedione diazonium solution prepared above was added dropwise. The reaction mixture was stirred at reflux for two additional hours, then allowed to cool as it stirred for about 18 hours. The reaction mixture was poured into about 600 mL of water, and the aqueous/organic layers were separated. The aqueous layer was washed with ethyl acetate, and the wash was combined with the organic layer. The combined organic material was washed with water, then with an aqueous solution saturated with sodium chloride. The organic material was dried with magnesium sulfate and filtered. The filtrate was concentrated under reduced pressure, yielding impure product. The product was dissolved in diethyl ether and washed with aqueous 10% hydrochloric acid, and with water. The diethyl ether solution was dried with magnesium sulfate and filtered. The filtrate was concentrated under reduced pressure, yielding 7.6 grams of title compound. The NMR spectrum was consistent with the proposed structure.

Step B 1-methyl-6-trifluoromethyl-3-(4-chloro-2-fluoro-5-hydroxy-6-nitrophenyl)-2,4(1H,3H)-pyrimidinedione

This compound was prepared in the manner of Step A of Example 1, using 3.8 grams (0.011 mole) of 1-methyl-6-trifluoromethyl-3-(4-chloro-2-fluoro-5-hydroxyphenyl)-2,4(1H,3H)-pyrimidinedione, 1.0 gram (0.011 mole) of 70% nitric acid, and 50 mL of sulfuric acid, yielding 1.5 grams of title compound. The NMR spectrum was consistent with the proposed structure.

Step C 1-methyl-6-trifluoromethyl-3-(6-amino-4-chloro-2-fluoro-5-hydroxy-phenyl)-2,4(1H,3H)-pyrimidinedione

This compound was prepared in the manner of Step B of Example 1, using 1.5 grams (0.004 mole) 1-methyl-6-trifluoromethyl-3-(4-chloro-2-fluoro-5-hydroxy-6-nitrophenyl)-2,4(1H,3H)-pyrimidinedione, 3.0 grams (0.054 mole) of iron dust, and 5 mL of water in 50 mL of glacial acetic acid, yielding 1.0 gram of title compound. The NMR spectrum was consistent with the proposed structure.

Step D Compound 28

This compound was prepared in the manner of Step C of Example 1, using 0.52 gram (0.0015 mole) of 1-methyl-6-trifluoromethyl-3-(6-amino-4-chloro-2-fluoro-5-hydroxyphenyl)-2,4(1H,3H)-pyrimidinedione, 0.18 gram (0.0017 mole) of isobutyryl chloride, 0.24 gram (0.0017 mole) of triethylamine, and 0.09 gram (0.0004 mole) of pyridinium p-toluenesulfonate in 50 mL of xylene, yielding 0.22 gram of Compound 28. The NMR spectrum was consistent with the proposed structure.

EXAMPLE 4

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SYNTHESIS OF 3-(4-CHLORO-6-FLUORO-2-PHENYLBENZOFURAN-7-YL)-1-METHYL-6-TRIFLUOROMETHYL-2,4(1H,3H)-PYRIMIDINEDIONE (Compound 280)

Step A ethyl N-(4-chloro-2,6-difluoro-3-nitrophenyl)carbamate

A stirred solution of 23.6 grams (0.109 mole) of ethyl N-(4-chloro-2,6-difluorophenyl)carbamate in 125 mL of concentrated sulfuric acid was cooled to about 0 °C and 7.7 mL (0.123 mole) of 70% nitric acid was added dropwise at a rate to maintain the reaction temperature below 10 °C. Upon completion of addition, the reaction mixture was stirred at 10 °C for 30 minutes and then allowed to warm to ambient temperature where it stirred for about 18 hours. At the conclusion of this period, the reaction mixture was poured into 150 mL of ice-water. The resulting precipitate was collected by vacuum filtration and washed with water followed by petroleum ether. The precipitate was dried in a heated vacuum desicator, yielding 30.6 grams of title compound. The NMR spectrum was consistent with the proposed structure.

Step B ethyl N-(4-chloro-6-fluoro-2-methoxy-3-nitrophenyl)carbamate

Under a nitrogen atmosphere, a solution of 30.6 grams (0.109 mole) of
ethyl N-(4-chloro-2,6-difluoro-3-nitrophenyl)carbamate and 18 mL (0.449 mole) of
methanol in 175 mL of dioxane was stirred and 218 mL (0.218 mole) of 1M sodium
trimethylsilanoate (in tetrahydrofuran) was added dropwise during a 45 minute period.

Upon completion of addition, the reaction mixture was heated to 65 °C where it stirred
for three hours. At the conclusion of this period, the reaction mixture was allowed to

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cool to ambient temperature where it stirred for about 18 hours. The reaction mixture was concentrated under reduced pressure to a residue. The residue was taken up in cold 3N hydrochloric acid. The resulting solid was collected by filtration, washed with petroleum ether, and heat dried under vacuum, yielding 21.3 grams of title compound. The NMR spectrum was consistent with the proposed structure.

Step C ethyl N-(3-amino-4-chloro-6-fluoro-2-methoxyphenyl)carbamate

Under a nitrogen atmosphere, a stirred solution of 21.3 grams (0.072 mole) of ethyl N-(4-chloro-6-fluoro-2-methoxy-3-nitrophenyl)carbamate, 18.3 grams (0.328 mole) of iron powder, 50 mL of acetic acid, and 250 mL of ethanol was heated to 65° C where it stirred for two hours. At the conclusion of this time, 3 mL (0.036 mole) of 12M hydrochloric acid was added. Upon completion of addition, the reaction mixture was stirred for an additional two hours. After this time, the reaction mixture was concentrated under reduced pressure to yield a brown oil. The oil was then taken up in methylene chloride. The mixture was filtered through diatomaceous earth, and the filter cake was washed with water and an aqueous saturated sodium bicarbonate solution. The filtrate was stored over sodium sulfate for about 18 hours and then filtered. The solvent was removed under reduced pressure to yield a black oil. This oil was filtered through a silica gel pad, yielding 15.0 grams of ethyl N-(3-amino-4-chloro-6-fluoro-2-methoxyphenyl)carbamate. The NMR spectrum was consistent with the proposed structure.

Step D 3-(3-amino-4-chloro-6-fluoro-2-methoxyphenyl)-6-trifluoromethyl-2,4-(1H,3H)-pyrimidinedione

This compound was prepared using 4.0 grams (0.036 mole) of sodium trimethylsilanolate, 6.6 grams (0.036 mole) of ethyl 3-amino-4,4,4-trifluorocrotonate, 8.5 grams (0.032 mole) of ethyl N-(3-amino-4-chloro-6-fluoro-2-methoxyphenyl)carbamate, and 2.2 grams (0.014 mole) of DBU in 75 mL of DMF. This preparation differs from well-known literature preparations for pyrimidinedione rings in that sodium trimethylsilanolate and DBU were used rather than sodium hydride. The yield of title compound was 1.7 grams. The NMR spectrum was consistent with the proposed structure.

Step E 3-(3-amino-4-chloro-6-fluoro-2-methoxyphenyl)-1-methyl-6trifluoromethyl-2,4(1H,3H)-pyrimidinedione

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A solution of 7.5 grams (0.021 mole) of 3-(3-amino-4-chloro-6-fluoro-2-methoxyphenyl)-6-trifluoromethyl-2,4-(1H,3H)-pyrimidinedione, 3.4 grams (0.025 mole) of potassium carbonate, and 3.5 grams (0.025 mole) of methyl iodide in 200 mL of acetone was stirred at ambient temperature for about 18 hours. The reaction mixture was then concentrated under reduced pressure, and the residue was taken up in 200 mL of water. The mixture was extracted with two 100 mL portions of ethyl acetate. The combined extracts were washed with two 50 mL portions of an aqueous saturated sodium chloride solution. The organic layer was dried with magnesium sulfate, filtered, and concentrated under reduced pressure, yielding 6.9 grams of crude product. The dark oil was combined with 7.0 grams of crude product prepared by a similar route to yield a total of 13.9 grams of crude product. The crude product was purified by column chromatography on silica gel, yielding 10.0 grams of title compound. The NMR spectrum was consistent with the proposed structure.

Step F 3-(4-chloro-6-fluoro-3-iodo-2-methoxyphenyl)-1-methyl-6-trifluoromethyl-2,4(1H,3H)-pyrimidinedione

A solution of 4.0 grams (0.011 mole) of 3-(3-amino-4-chloro-6-fluoro-2-methoxyphenyl)-1-methyl-6-trifluoromethyl-2,4-(1H,3H)-pyrimidinedione in 25 mL (0.300 mole) of concentrated hydrochloric acid was stirred and cooled in an ice bath. During a 15 minute period, 1.9 grams (0.013 mole) of sodium nitrite was added dropwise at a rate to maintain the reaction temperature at 15 °C. Upon completion of addition, the mixture was stirred for 20 minutes and then poured into 15.0 grams (0.090 mole) of potassium iodide. The reaction mixture was stirred for 30 minutes and then filtered. The filter cake was thoroughly washed with distilled water and then taken up in 150 mL of ethyl acetate. The resulting solution was dried with sodium sulfate and filtered. The filtrate was concentrated under reduced pressure to yield a brown solid. The solid was subjected to column chromatography on silica gel. Elution was accomplished using 5:1 heptane and ethyl acetate. The product-containing fractions were combined and concentrated under reduced pressure, yielding 3.0 grams of title compound. The NMR spectrum was consistent with the proposed structure.

Step G 3-(4-chloro-6-fluoro-2-hydroxy-3-iodophenyl)-1-methyl-6trifluoromethyl-2,4(1H,3H)-pyrimidinedione

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Under a nitrogen atmosphere, a stirred solution of 3.0 grams (0.006 mole) of 3-(4-chloro-6-fluoro-3-iodo-2-methoxyphenyl)-1-methyl-6-trifluoromethyl-2,4(1H,3H)-pyrimidinedione in 75 mL of methylene chloride was cooled in a dry ice/acetone bath and 22.0 mL (0.022 mole) of 1M boron tribromide (in methylene chloride) was added dropwise during a 20 minute period. Upon completion of addition, the reaction mixture was allowed to warm to ambient temperature were it stirred for about one hour. At the conclusion of this period, the reaction mixture was poured into 200 mL of water and extracted with two 50 mL portions of methylene chloride. The combined extracts were washed with one 100 mL portion of an aqueous saturated sodium chloride solution, dried with sodium sulfate, and filtered. The filtrate was concentrated under reduced pressure, yielding 2.6 grams of title compound. The NMR spectrum was consistent with the proposed structure.

Step H Compound 280

Under a nitrogen atmosphere, a solution of 1.5 grams (0.003 mole) of 3-(4-chloro-6-fluoro-2-hydroxy-3-iodophenyl)-1-methyl-6-trifluoromethyl-2,4(1H,3H)pyrimidinedione, 0.41 gram (0.004 mole) of phenylacetylene, and 0.71 gram (0.007 mole) of triethylamine in 25 mL of DMF was stirred. To this was added 0.09 gram (0.00013 mole) of dichlorobis(triphenylphosphine)pallidium (II) and 0.05 gram (0.00026 mole) of copper (I) iodide. Upon completion of addition, the reaction mixture was heated to 70 °C where it stirred for 2.5 hours. After this time, the reaction mixture was cooled to ambient temperature and then poured into 150 mL of an aqueous 10% ammonium chloride solution. The resulting precipitate was collected by filtration and washed with water. The precipitate was taken up in 120 mL of ethyl acetate. The resulting solution was dried with sodium sulfate and filtered. The filtrate was concentrated under reduced pressure to a brown solid. The solid was recrystallized using 1:1 chloroform and petroleum ether, yielding 0.31 gram of Compound 280. The mother liquor was concentrated to a residue. The residue was recrystallized using petroleum ether to yield an additional 0.21 gram of Compound 280, m.p. 215-216 °C. The NMR spectrum was consistent with the proposed structure.

EXAMPLE 5

SYNTHESIS OF 3-(4-CHLORO-6-FLUORO-2-TRIFLUOROMETHYLBENZIMIDAZOL-7-YL)-1-METHYL-6-TRIFLUOROMETHYL-2,4(1H,3H)-PYRIMIDINEDIONE

5 (Compound 365)

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A stirred solution of 3.0 grams (0.0085 mole) of 3-(5,6-diamino-4-chloro-2-fluorophenyl)-1-methyl-6-trifluoromethyl-2,4(1H,3H)-pyrimidinedione in 15.0 mL of trifluoroacetic acid was heated to 65 °C where it stirred for one hour. At the conclusion of this period, the reaction mixture was analyzed by TLC, which indicated that the reaction was not complete. The reaction mixture was stirred at 65 °C for an additional two hours. After this time, the reaction mixture was again analyzed by TLC, which indicated that the reaction was complete. The reaction mixture was allowed to cool to ambient temperature and then poured into 200 mL of water. The resulting mixture was allowed to stand at ambient temperature for about 18 hours. At the conclusion of this period, the resulting solid was collected by filtration and washed with water followed by heptane. The filter cake was dried under vacuum, yielding 3.6 grams of Compound 365, m.p. 130 °C. The NMR spectrum was consistent with the proposed structure.

EXAMPLE 6

20 SYNTHESIS OF 3-(4-CHLORO-2-ETHYL-6-FLUOROBENZIMIDAZOL-7-YL)-1-METHYL-6-TRIFLUOROMETHYL-2,4(1H,3H)-PYRIMIDINEDIONE (COMPOUND 367)

Step A 3-(4-chloro-2,6-difluorophenyl)-1-methyl-6-trifluoromethyl-2,4-(1H,3H)-pyrimidinedione

Under a nitrogen atmosphere, a solution of 32.0 grams (0.900 mole) of sodium hydride (60% by weight) in 250 mL of DMF was vigorously stirred and cooled in an ice bath. To this a solution of 133.0 grams (0.726 mole) of ethyl 3-amino-4,4,4-trifluorocrotonate in 150 mL of DMF was added dropwise at a rate to maintain the reaction mixture temperature at about 5 °C. Upon completion of addition, a solution of 156.3 grams (0.663 mole) of ethyl N-(4-chloro-2,6-difluorophenyl)carbamate in 250 mL of DMF was added dropwise. Upon completion of addition, the mixture was

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removed from the ice bath and heated to 130 °C where it stirred for 3.5 hours. After this time, the mixture was analyzed by gas chromatography (GC), which indicated that only a slight amount of the starting material was left. The mixture was cooled to 5 °C and 83.0 mL (1.333 moles) of methyl iodide was added dropwise at a rate to maintain the reaction mixture temperature below 20 °C. Upon completion of addition, the reaction mixture was allowed to warm to ambient temperature where it stirred for about 18 hours. At the conclusion of this period, the reaction mixture was filtered through diatomaceous earth. The filtrate was concentrated under reduced pressure to yield a dark viscous oil. The oil was taken up in methylene chloride and washed with three 1000 mL portions of water followed by one 1000 mL portion of an aqueous saturated sodium chloride solution. The organic layer was dried with magnesium sulfate, filtered, and concentrated under reduced pressure, yielding 223.8 grams of title compound. The NMR spectrum was consistent with the proposed structure.

Step B 3-(4-chloro-2,6-dif]uoro-5-nitrophenyl)-1-methyl-6-trifluoromethyl-2,4(1H,3H)-pyrimidinedione

A stirred solution of 211.0 grams (0.619 mole) of 3-(4-chloro-2,6difluorophenyl)-1-methyl-6-trifluoromethyl-2,4(1H,3H)-pyrimidinedione in 600 mL of concentrated sulfuric acid was cooled to less than 10 °C, and 44 mL (0.689 mole) of aqueous 70% nitric acid was added dropwise at a rate to maintain the reaction temperature below 10 °C. Upon completion of addition, the reaction mixture was analyzed by GC, which indicated the reaction was incomplete. The reaction was allowed to warm to ambient temperature and an additional 5 mL (0.078 mole) of aqueous 70% nitric acid was added. The reaction mixture was again analyzed by GC, which indicated the reaction was complete. The reaction mixture was poured into ice-water. The resulting solid was collected by filtration, washed with water, and then taken up in 600 mL of methylene chloride. The resulting solution was washed with two 600 mL portions of water, one 600 mL portion of an aqueous saturated sodium bicarbonate solution, and one 600 mL portion of an aqueous saturated sodium chloride solution. The organic layer was separated, dried with magnesium sulfate, and filtered. The filtrate was concentrated under reduced pressure, yielding a waxy tan solid. The solid was triturated with heptane and allowed to stand for about 72 hours. At the conclusion of this period, the solid was collected by filtration,

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washed with heptane, and dried under reduced pressure, yielding 201.4 grams of title compound. The NMR spectrum was consistent with the proposed structure.

Step C 3-(6-amino-4-chloro-2-fluoro-5-nitrophenyl)-1-methyl-6-trifluoromethyl-2,4(1H,3H)-pyrimidinedione

To stirred solution of 200 grams (0.519 mole) of 3-(4-chloro-2,6-difluoro-5-nitrophenyl)-1-methyl-6-trifluoromethyl-2,4(1H,3H)-pyrimidinedione in 1000 mL of dioxane was added 150 mL (1.091 moles) of triethylamine in one portion. Upon completion of addition, the mixture was vigorously stirred and 400 grams (5.189 moles) of ammonium acetate was added in one portion. The reaction mixture was heated to 90 °C where it stirred for two hours. The reaction mixture was allowed to cool to ambient temperature where it stirred for about 18 hours. The resulting suspension was collected by filtration and washed with dioxane. The filtrate was concentrated under reduced pressure to yield a viscous dark oil. The oil was poured into ice-water. The resulting solid was collected by filtration and washed with water. The solid was dried under reduced pressure and then at ambient temperature for about 18 hours, yielding 195.1 grams of title compound. The NMR spectrum was consistent with the proposed structure.

Step D 3-(5,6-diamino-4-chloro-2-fluorophenyl)-1-methyl-6-trifluoromethyl-2,4-(1H,3H)-pyrimidinedione and 3-(5,6-diamino-4-chloro-2-fluorophenyl)-1-methyl-6-trifluoromethyl-2,4(1H,3H)-pyrimidinedione

A solution of 278.0 grams (1.232.moles) of tin(II) chloride dihydrate, 264.0 grams (4.936 moles) of ammonium chloride, 400 mL of water. and 800 mL of ethanol was vigorously stirred, and 157.4 grams (0.411 mole) of 3-(6-amino-4-chloro-2-fluoro-5-nitrophenyl)-1-methyl-6-trifluoromethyl-2,4(1H,3H)-pyrimidinedione was added. Upon completion of addition, the reaction mixture was heated to 83-85 °C where it stirred for 18 hours. After this time the reaction mixture was allowed to cool to ambient temperature. The resultant solid by-product was collected by filtration and washed with ethanol. The combined filtrate and wash was concentrated under reduced pressure to yield a suspension of additional by-product. The suspension was taken up in ethyl acetate and the resultant emulsion was filtered through a pad of diatomaceous earth. The filter cake was washed with ethyl acetate, and the combined organics were washed with three 200 mL portions of water. The organic

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layer was dried with magnesium sulfate, filtered, and concentrated under reduced pressure to a brown residue. The residue was triturated with heptane and allowed to stand for about five days. The resultant solid was collected by filtration and dried, yielding 144.4 grams of crude product. The crude product was combined with material prepared by a similar route, yielding a total of 157.8 grams of material. The combined product was subjected to column chromatography on silica gel, yielding 83.2 grams of an orange solid. The solid was slurried with warm ethyl acetate, and the insoluble product was collected by filtration. The product was washed with ethyl acetate, and the wash and filtrate from above were combined. The process of concentrating the filtrate, and slurrying the solid residue was repeated twice more, yielding a total of 51.9 grams of title compound. The NMR spectrum was consistent with the proposed structure.

An alternate method for preparing 3-(5,6-diamino-4-chloro-2-fluorophenyl)-1-methyl-6-trifluoromethyl-2,4(1H,3H)-pyrimidinedione is the following:

A solution of 19.2 grams (0.050 mole) of 3-(6-amino-4-chloro-2-fluoro-5-nitrophenyl)-1-methyl-6-trifluoromethyl-2,4(1H,3H)-pyrimidinedione, 3.0 grams (0.056 mole) of ammonium chloride, and 50 mL of water in 100 mL of ethanol was stirred, and 11.2 grams (0.201 mole) of iron powder (325 mesh) was added in one portion. Upon completion of addition, the reaction mixture was heated at reflux for one hour. The reaction mixture was allowed to cool to ambient temperature, then it was filtered through diatomaceous earth to remove the iron powder. The filter cake was washed with 200 mL of acetone, and the wash was combined with the filtrate. The combination was stirred with decolorizing carbon and filtered. The filtrate was concentrated under reduced pressure, yielding a dark brown oil. The oil was then taken up in 200 mL of methylene chloride and washed with three 100 mL portions of an aqueous saturated sodium bicarbonate solution. The organic layer was dried with magnesium sulfate, filtered, and concentrated under reduced pressure, yielding 12.8 grams of title compound. The NMR spectrum was consistent with the proposed structure.

30 Step E Compound 367

A stirred solution of 1.0 grams (0.0028 mole) of 3-(5,6-diamino-4-chloro-2-fluorophenyl)-1-methyl-6-trifluoromethyl-2,4(1H,3H)-pyrimidinedione and 0.28 mL

(0.0035 mole) of pyridine in 10 mL chloroform was cooled to 5 °C and 0.27 mL (0.0031 mole) of propionyl chloride was added dropwise. Upon completion of addition, the mixture was allowed to warm to ambient temperature were it stirred for about 18 hours. The mixture was cooled to 5 °C and 5.0 mL (0.054 mole) of phosphorous oxychloride was added in one portion. Upon completion of addition, the reaction mixture was allowed to warm to ambient temperature where it stirred for about 18 hours. At the conclusion of this period, the reaction mixture was poured into 200 mL of cold water, the resulting mixture was stirred for one hour, then it was extracted with three 50 mL portions of chloroform. The combined extracts were dried with magnesium sulfate and filtered. The filtrate was concentrated under reduced pressure, yielding 0.15 gram of an orange residue. The aqueous layer was made basic with an aqueous saturated sodium bicarbonate solution to a pH of 3-4. The resulting mixture was extracted with three 50 mL portions of methylene chloride. The extracts were combined, dried with magnesium sulfate, and filtered. The filtrate was concentrated under reduced pressure, yielding 0.70 gram of a yellow residue. The yellow residue was triturated with hot heptane. The resulting solid was collected by filtration and washed with heptane, yielding 0.67 gram of Compound 367, m.p. 150-155 °C. The NMR spectrum was consistent with the proposed structure.

EXAMPLE 7

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SYNTHESIS OF 3-(2-T-BUTYL-4-CHLORO-6-FLUOROBENZIMIDAZOL-7-YL)-1-METHYL-6-TRIFLUOROMETHYL-2,4(1H,3H)-PYRIMIDINEDIONE (Compound 369)

To a stirred solution of 1.0 grams (0.0028 mole) of 3-(5,6-diamino-4-chloro-2-fluorophenyl)-1-methyl-6-trifluoromethyl-2,4(1H,3H)-pyrimidinedione , 15.0 mL of ethanol, and 4 mL of 5M hydrochloric acid was added 1.2 mL (0.0057 mole) of 2,2,6,6-tetramethyl-3,5-heptanedione. Upon completion of addition, the reaction mixture was heated to reflux where it stirred for ten minutes. At the conclusion of this period, the reaction mixture was analyzed by TLC, which indicated that the reaction was not complete. The reaction mixture was stirred at reflux for an additional two hours. After this time, the reaction mixture was again analyzed by TLC, which again indicated that the reaction was still not complete. As a result, an additional 1.0 mL

(0.0048 mole) of 2,2,6,6-tetramethyl-3,5-heptanedione was added. Upon completion of addition, the reaction mixture was stirred at reflux for three days. At the conclusion of this period, more ethanol was added to replace that which evaporated, and the reaction mixture was analyzed by TLC for a third time. The reaction mixture was allowed to cool to ambient temperature, poured into 100 mL of an aqueous saturated sodium bicarbonate solution, and 100 mL of chloroform was added. The aqueous layer was separated and washed with two 100 mL portions of chloroform. The chloroform layer and washes were combined, dried with magnesium sulfate, and filtered. The filtrate was treated with decolorizing carbon and stirred. The mixture was filtered and concentrated under reduced pressure to yield a red oil. The oil was taken up in heptane. The resulting solid was collected by filtration and washed with heptane to yield a tan solid. The solid was purified by column chromatography on silica gel, yielding 0.36 gram of Compound 369, m.p. 125-130 °C. The NMR spectrum was consistent with the proposed structure.

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EXAMPLE 8

SYNTHESIS OF 3-(7-CHLORO-5-FLUORO-2-TRIFLUOROMETHYLINDOL-4-YL)-1-METHYL-6-TRIFLUOROMETHYL-2,4(1H,3H)-PYRIMIDINEDIONE (Compound 500)

20 Step A 3-[5-(1-trifluoromethylethylidenehydrazino)-4-chloro-2-fluorophenyl]-1-methyl-6-trifluoromethyl-2,4(1H,3H)-pyrimidinedione

A solution of 3.37 grams (0.010 mole) of 3-(5-amino-4-chloro-2-fluorophenyl)-1-methyl-6-trifluoromethyl-2,4(1H,3H)-pyrimidinedione in 80 mL of concentrated hydrochloric acid was stirred at 25 °C for 20 minutes. After this time, the solution was cooled to 10 °C and a solution of 0.69 gram (0.010 mole) of sodium nitrite in 10 mL of water was slowly added. Upon completion of addition, the mixture was stirred for one hour at 10 °C and then a solution of 5.64 grams (0.025 mole) of tin (II) chloride dihydrate in 40 mL of concentrated hydrochloric acid was slowly added. Upon completion of addition, the reaction mixture was warmed to 25 °C where it stirred for one hour. At the conclusion of this period, 1.12 grams (0.010 mole) of trifluoroacetone was added and the resulting solid was collected by filtration,

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yielding 3.13 grams of title compound, m.p. 213-214 °C. The NMR spectrum was consistent with the proposed structure.

Step B Compound 500

A stirred solution of 2.0 grams (0.0044 mole) of 3-[5-(1-trifluoromethylethylidenehydrazino)-4-chloro-2-fluorophenyl]-1-methyl-6-trifluoromethyl-2,4(1H,3H)-pyrimidinedione in 80 mL of polyphosphoric acid was heated at 80 °C for 20 minutes. After this time, the reaction mixture was allowed to cool to 25 °C where it was diluted with water. The resulting solid was collected by filtration, yielding 0.73 gram of Compound 500, m.p. 208-210 °C. The NMR spectrum was consistent with the proposed structure.

EXAMPLE 9

SYNTHESIS OF 3-(7-CHLORO-2-ETHOXYCARBONYLINDOL-4-YL)-4,5,6,7-TETRAHYDRO-1H-ISOINDOLE-1,3(2H)-DIONE (Compound 595)

This compound was prepared in the manner of Step A, Example 1, using, 17.25 grams (0.10 mole) of 2-chloro-5-nitroaniline, 6.9 grams (0.10 mole) of sodium nitrite, 56.4 grams (0.25 mole) of tin (II) chloride dihydrate, 11.61 grams (0.10 mole) of ethyl pyruvate, 30 mL of water, and 100 mL of concentrated hydrochloric acid. This preparation differs in that ethyl pyruvate was used rather than trifluoroacetone. The yield of title compound was 19.4 grams. The NMR spectrum

Step B 7-chloro-2-ethoxycarbonyl-4-nitroindole

was consistent with the proposed structure.

This compound was prepared in the manner of Step B, Example 8, using 14.0 grams (0.050 mole) of 3-(1-ethoxycarbonylethylidenehydrazino)-4-chloronitrobenzene in 100 mL of polyphosphoric acid. The yield of title compound was 0.4 gram. The NMR spectrum was consistent with the proposed structure.

Step C 7-amino-4-chloro-2-ethoxycarbonylindole

A stirred solution of 2.68 grams (0.01 mole) of 4-chloro-2-30 ethoxycarbonyl-7-nitroindole, 80 mL of acetic acid, and 15 mL of water was heated to 65 °C, and 18.3 grams (0.048 mole) of iron powder was slowly added during a 20

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minute period. Upon completion of addition, the reaction mixture was allowed to cool to 25 °C where it stirred for one hour. After this time, the reaction mixture was poured into water, and the resulting mixture was filtered through diatomaceous earth. The filter cake was washed thoroughly with ethyl acetate. The organic layer was dried with magnesium sulfate and filtered. The filtrate was concentrated under reduced pressure a residue. The residue was purified by column chromatography, yielding 0.4 gram of title compound. The NMR spectrum was consistent with the proposed structure.

Step D Compound 595

A stirred solution of 0.4 gram (0.0016 mole) of 7-amino-4-chloro-2-ethoxycarbonylindole and 0.26 gram (0.0016 mole) of 3,4,5,6-tetrahydrophhalic anhydride in 80 mL of acetic acid was heated at reflux for about 18 hours. After this time, the reaction mixture was extracted with several portions of diethyl ether. The organic extracts were combined, dried with magnesium sulfate, and filtered. The filtrate was concentrated under reduced pressure to a residue. The residue was purified by column chromatography on silica gel, yielding 0.47 gram of Compound 595. The NMR spectrum was consistent with the proposed structure.

PCT/US98/03647

_- 28 -

<u>Table 1</u> Benzoxazoles

where A is nitrogen double bonded to position 2 and B is O; J is

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	Compound No.	X	<u>R</u>	<u>R3</u>
10	1 2 3 4 5	4-Cl, 6-F 4-Cl, 6-F 4-Cl, 6-F 4-Cl, 6-F	CH ₃ CH ₃ CH ₃ CH ₃	${ m CH_3} \ { m C_2H_5} \ { m CH_2CN} \ { m CH_2CH=CH_2} \ { m NH_2} \ { m CH_2C\equiv CH}$
15	6 7 8 9	4-CI, 6-F 4-CI, 6-F 4-CI, 6-F 4-CI, 6-F	CH₃ CH₃ CH₃ CH₃	$CH_2C \equiv CH$ C_3H_7 CH_2OCH_3 $CH_2CO_2C_2H_5$

Table 2

$$X(n) \xrightarrow{A} B \\ 2 R$$

$$X(n) \xrightarrow{A} B \\ 2 R$$

$$0 \xrightarrow{N-CH_3} 0 \xrightarrow{N-CH_3} 0 \xrightarrow{N-NH_2} 0 \xrightarrow{N-N} 0$$

$$CF_3 CF_3 CF_3 CF_3$$

$$J1 J2 J3 J4 J5$$

$$0 \xrightarrow{N-CHF_2} 0$$

$$CH_3 0 \xrightarrow{N-CHF_2} 0$$

$$J6 J7$$

	<u>No.</u>	A	<u>B</u>	Double Bond Posit'n	X	<u>R</u>	ī
	10	N	0	1-2	4-CI	CH ₃	J1
10	11	N	0	1-2	4-CI	C_2H_5	J1
	12	N	0	1-2	4-Cl	CH(CH ₃) ₂	J1
	13	N	О	1-2	4,6-Cl ₂	CH₃	J1
	14	N	0	1-2	4,6-Cl ₂	C ₂ H ₅	J1
	15	N	0	1-2	4,6-Cl ₂	C₂H₅	J1
15	16	N	Ο	1-2	4-Br, 6-F	CH₃	J1
	17	N	0	1-2	4-CF ₃ ,6-F	CH₃	J1
	18	N	О	1-2	4,6-F ₂	CH₃	J1
	19	N	0	1-2	4-CN, 6-F	CH₃	J1
	20	N	0	1-2	4-OCF ₃ ,6-F	CH₃	J1
20	21	N	0	1-2	4-Br, 6-F	C₂H₅	J1
	22	N	Ο	1-2	4-CN, 6-F	C₂H₅	J1
	2 3	N	0	1-2	4-CN, 6-F	CH(CH ₃) ₂	J1
	24	N	Ο	1-2	4-CH ₃ , 6-F	CH₃	J1
	25	N	0	1-2	4-CI, 6-F	C_2H_5	J1
25	26	N	Ο	1-2	4-CI, 6-F	C₃H ₇	J1
	27	N	0	1-2	4-CI, 6-F	C₄H ₉	J1
	28	N	0	1-2	4-CI, 6-F	CH(CH ₃) ₂	J1
	29	N	0	1-2	4-CI, 6-F	CH ₂ CH(CH ₃) ₂	J1

	30	N	0	1-2	4-Cl, 6-F	C(CH ₃) ₃	J1
	31	N	0	1-2	4-CI, 6-F	phenyl -	J1
	32	N	0	1-2	4-CI, 6-F	phenylmethyl	J1
	33	N	0	1-2	4-Cl, 6-F	CF ₃	J1
5	34	N	´- o	1-2	4-Cl, 6-F	CCl ₂	J1
	3 5	N	0	1-2	4-Cl, 6-F	CI	J1
	36	N	0	1-2	4-CI, 6-F	ОН	J1
	37	N	0	1-2	4-CI, 6-F	Br	J1
	38	N	0	1-2	4-CI, 6-F	NH ₂	J1
10	3 9	N	0	1-2	4-Cl, 6-F	NHCH₃	J1
	40	N	0	1-2	4-CI, 6-F	N(CH ₃) ₂	J1
	41	N	0	1-2	4-CI, 6-F	NHCH ₂ CO ₂ CH ₃	J1
	42	Ν	0	1-2	4-CI, 6-F	NHSO ₂ CH ₃	J1
	43	Ν	0	1-2	4-Br, 6-F	NHCOCH₃	J1
15	44	Ν	0	1-2	4-CI, 6-F	morpholino	J1
	45	Ν	0	1-2	4-CI, 6-F	NHSO ₂ C ₆ H ₅	J1
	4 6	N	0	1-2	4-Cl, 6-F	NHSO ₂ CH ₂ C ₆ H ₅	J1
	47	N	0	1-2	4-CI, 6-F	N(CH ₃)SO ₂ CH ₃	J1
	4 8	N	0	1-2	4-Cl, 6-F	NHPO(OCH ₃) ₂	J1
20	49	N	0	1-2	4-Br, 6-F	CH ₂ CO ₂ CH ₃	J1
	50	N	0	1-2	4-CI, 6-F	C₂H₄CO₂CH₃	J1
	51	N	0	1-2	4-CI, 6-F	CH=CHCO₂CH₃	J1
	52	Ν	0	1-2	4-CI, 6-F	CH=C(CI)CO ₂ CH ₃	J1
	5 3	N	0	1-2	4-CI, 6-F	CH ₂ CH(CI)CO ₂ CH ₃	J1
25	54	Ν	0	1-2	4-CI, 6-F	OCH₃	J1
	5 5	Ν	0	1-2	4-CI, 6-F	OC₂H₅	J1
	56	Ν	0	1-2	4-Cl, 6-F	OCH(CH ₃) ₂	J1
	57	Ν	0	1-2	4-CI, 6-F	OCH ₂ CH=CH ₂	J1
	5 8	N	0	1-2	4-CI, 6-F	OCH ₂ C(CH ₃)=CH ₂	J1
30	5 9	Ν	О	1-2	4-CI, 6-F	OCH₂ CC H	J1
	6 0	Ν	О	1-2	4-CI, 6-F	OCH ₂ CO ₂ C ₂ H ₅	J1
	61	N	0	1-2	4-CI, 6-F	OCH(CH ₃)CO ₂ CH ₃	J1
	62	N	0	1-2	4-CI, 6-F	OCH₂CN	J1
	6 3	N	Ο	1-2	4-CI, 6-F	OCH ₂ CONH ₂	J1
35	64	N	О	1-2	4-CI, 6-F	OCH ₂ CONHCH ₃	J1
	65	N	Ο	1-2	4-CI, 6-F	OCH(CH ₃)CONH ₂	J1
	6 6	Ν	0	1-2	4-CI, 6-F	OCH(CH ₃)CONHCH ₃	J1
	67	N	Ο	1-2	4-CI, 6-F	OCH ₂ CO ₂ H	J1
	68	Ν	Ο	1-2	4-Cl, 6-F	phenoxy	J1
40	6 9	Ν	Ο	1-2	4-CI, 6-F	p-OC ₆ H ₄ OCH(CH ₃)CO ₂ CH ₃	J1
	70	. N	0	1-2	4-CI, 6-F	4-chlorophenoxy	J1
	71	Ν	0	1-2	4-CI, 6-F	phenylmethoxy	J1
	72	Ν	0	1-2	4-Ci, 6-F	CN	J1
	73	N	Ο	1-2	4-CI, 6-F	CO ₂ CH ₃	J1
45	74	N	0	1-2	4-CI, 6-F	CO₂H	J1
	75	N	0	1-2	4-CI, 6-F	CO₂Na	J1
	76	N	0	1-2	4-CI, 6-F	CONH ₂	J1
	77	Ν	0	1-2	4-CI, 6-F	CONHCH ₃	J1
	78	Ν	0	1-2	4-CI, 6-F	CON(CH ₃) ₂	J1
50	79	N	0	1-2	4-CI, 6-F	CONHSO₂CH₃	J1

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.- 31 -

	80	N	Ο	1-2	4-CI, 6-F	CO₂NHOCH₃	J1
	81	N	0	1-2	4-CI, 6-F	SCH₃	J1
	82	Ν	0	1-2	4-Cl, 6-F	SCH ₂ CO ₂ CH ₃	J1
	83	N	. 0	1-2	4-CI, 6-F	SCH₂CONH₂	J1
5	84	N	- o	1-2	4-CI, 6-F	SO₂CH₃	J1
_	'8 5	N	0	1-2	4-CI, 6-F	SH	J1
	8 6	N	0	1-2	4-CI, 6-F	CH₂OH	J1
	87	N	0	1-2	4-CI, 6-F	CH(CH₃)OH	J1
	88	N	0	1-2	4-CI, 6-F	C(CH ₃) ₂ OH	J1
10	8 9	N	0	1-2	4-CI, 6-F	C₂H₄OH	J1
	90	N	0	1-2	4-CI, 6-F	CH₂CH(CH₃)OH	J1
	91	N	0	1-2	4-CI, 6-F	CH ₂ C(CH ₃) ₂ OH	J1
	92	N	0	1-2	4-CI, 6-F	C(CH ₃) ₂ OCOCH ₃	J1
	93	N	0	1-2	4-CI, 6-F	CH(CH ₃) ₂ OCOCH ₃	J1
15	94	N	0	1-2	4-CI, 6-F	CH(CH ₃)OCOCH ₃	J1
	95	N	0	1-2	4-CI, 6-F	CHBr ₂	J1
	96	N	0	1-2	4-Br, 6-F	CH₂OCH₃	J1
	97	N	0	1-2	4-CI, 6-F	CH₂OCH₂CCH	J1
	98	N	Ο	1-2	4-Br, 6-F	NH ₂	J1
20	99	N	0	1-2	4-Br, 6-F	phenoxymethyl	J1
	100	N	0	1-2	4-Br, 6-F	N(COCH ₃) ₂	J1
	101	N	0	1-2	4-Br, 6-F	CH₂OCOCH₃	J1
	102	N	0	1-2	4-Br, 6-F	4-chlorophenoxymethyl	J1
	103	N	0	1-2	4-Br, 6-F	CH(Ph)OCOCH₃	J1
2 5	104	N	0	1-2	4-Br, 6-F	C(CH ₃) ₂ OCOCH ₃	J1
	105	N	0	1-2	4-Br, 6-F	CO₂H	J1
	106	N	О	1-2	4-Br, 6-F	OCH₂ CC H	J1
	107	N	0	1-2	4-Br, 6-F	OCH(CH ₃) ₂	J1
	108	N	О	1-2	4-Br, 6-F	NHSO ₂ CH ₃	J1
30	109	N	0	1-2	4-Br, 6-F	OCH ₃	J1
•	110	N	0	1-2	4-Br, 6-F	OCH ₂ CH=CH ₂	J1
	111	N	0	1-2	4-CI, 6-F	(CH ₃)(CN)OH	J1
	112	N	0	1-2	4-Cl, 6-F	CH ₃	J2
	113	N	О	1-2	4-CI, 6-F	$n-C_3H_7$	J2
35	114	N	0	1-2	4-Cl, 6-F	i-C₃H ₇	J2
•	115	N	O	1-2	4-CI, 6-F	t-C₄H ₉	J2
	116	N	o	1-2	4-CI, 6-F	C ₂ H ₅	J2
	117	N	0	1-2	4-CI, 6-F	CH ₂ CO ₂ CH ₃	J2
	118	N	0	1-2	4-CI, 6-F	phenoxymethyl	J2
4 0	119	N	0	1-2	4-CI, 6-F	CONHCH ₃	J2
• •	120	N	0	1-2	4-CI, 6-F	CON(CH ₃) ₂	J2
	121	N	0	1-2	4-CI, 6-F	CO ₂ CH ₃	J2
	122	N	0	1-2	4-CI, 6-F	Phenyl	J2
	123	N	0	1-2	4-CI, 6-F	SCH ₃	J2
4 5	124	N	Ō	1-2	4-CI, 6-F	CH₂OCH₃	J2
47	125	N	0	1-2	4-CI, 6-F	Benzyl	J2
	126	N	0	1-2	4-CI, 6-F	4-chlorophenylmethyl	J2
	127	N	Ö	1-2	4-CI, 6-F	SO₂CH₃	J2
	128	N	0	1-2	4-CI, 6-F	CF ₃	J2
50	129	N	0	1-2	4-CI, 6-F	C(CH ₃) ₂ OCO ₂ CH ₃	J2
20	123	14	•	. –			

.- 32 -

			_	4.5	4 01 0 5	0/011 \ 011 011	J2
	130	N	0	1-2	4-Cl, 6-F	C(CH ₃) ₂ CH ₂ OH	J2 J3
	131	N	0	1-2	4-Cl, 6-F	CH₃	J3
	132	N	0	1-2	4-Cl, 6-F	n-C ₃ H ₇	J3
	133	N	. 0	1-2	4-Cl, 6-F	i-C ₃ H ₇	
5	134	N	0	1-2	4-Cl, 6-F	t-C₄H ₉	J3
	135	N	0	1-2	4-Cl, 6-F	CH₂OH	J3
	136	N	. O	1-2	4-Cl, 6-F	CH₂CH₂OH	J3
	137	N	O .	1-2	4-Cl, 6-F	C(CH ₃) ₂ OH	J3
	138	N	О	1-2	4-CI, 6-F	CONHCH₃	J3
10	139	Ν	0	1-2	4-CI, 6-F	CON(CH ₃) ₂	J3
	140	Ν	0	1-2	4-CI, 6-F	CO₂CH₃	J3
	141	N	0	1-2	4-CI, 6-F	Phenyl	J3
	142	Ν	0	1-2	4-Cl, 6-F	SCH₃	J3
	143	N	0	1-2	4-Cl, 6-F	CH₂OCH₃	J3
15	144	N	0	1-2	4-Cl, 6-F	Benzyl	J3
	145	Ν	0	1-2	4-CI, 6-F	4-chlorophenylmethyl	J3
	146	Ν	0	1-2	4-CI, 6-F	SO₂CH₃	J 3
	147	N	0	1-2	4-CI, 6-F	CF ₃	J3
	148	Ν	0	1-2	4-Cl, 6-F	$C(CH_3)_2OCO_2CH_3$	J3
20	149	N	0	1-2	4-CI, 6-F	C(CH ₃) ₂ CH ₂ OH	J3
	150	N	0	1-2	4-CI, 6-F	C(CH ₃) ₂ CH ₂ OCH ₃	J3
	151	N	0	1-2	4-CI, 6-F	C₂H₅	J3
	152	N	0	1-2	4-Cl, 6-F	CO₂Na	J3
	153	N	0	1-2	4-CI, 6-F	CONHSO₂CH₃	J 3
2 5	154	N	0	1-2	4-CI, 6-F	OCH ₂ CO ₂ CH ₃	J 3
	15 5	Ν	0	1-2	4-Cl, 6-F	OCH(CH ₃)CO ₂ CH ₃	J3
	156	N	0	1-2	4-CI, 6-F	OCH ₂ CH=CH ₂	J3
	157	N	0	1-2	4-CI, 6-F	OCH₂ CCH	J3
	158	N	0	1-2	4-CI, 6-F	ОН	J3
30	159	N	0	1-2	4-CI, 6-F	OCH ₃	JЗ
	160	N	0	1-2	4-Cl, 6-F	OCH(CH ₃) ₂	J3
	161	Ν	0	1-2	4-CI, 6-F	CH₃	J4
	162	N	О	1-2	4-CI, 6-F	n-C ₃ H ₇	J4
	163	N	О	1-2	4-CI, 6-F	i-C ₃ H ₇	J4
35	164	N	О	1-2	4-Cl, 6-F	t-C₄H ₉	J4
	165	N	О	1-2	4-CI, 6-F	CH₂OH	J4
	166	N	0	1-2	4-CI, 6-F	CH₂CH₂OH	J4
	167	N	О	1-2	4-Cl, 6-F	C(CH ₃) ₂ OH	J4
	168	N	0	1-2	4-CI, 6-F	CONHCH₃	J4
4 0	169	N	О	1-2	4-CI, 6-F	CON(CH ₃) ₂	J4
	170	N	О	1-2	4-CI, 6-F	CO ₂ CH ₃	J4
	171	N	0	1-2	4-CI, 6-F	Phenyl	J4
	172	N	О	1-2	4-Cl, 6-F	SCH ₃	J4
	173	N	0	1-2	4-CI, 6-F	CH ₂ OCH ₃	J4
45	174	N	0	1-2	4-Cl, 6-F	Benzyl	J4
	175	N	0	1-2	4-Cl, 6-F	4-chlorophenylmethyl	J4
	176	N	0	1-2	4-CI, 6-F	SO₂CH₃	J4
	177	N	0	1-2	4-Cl, 6-F	CF ₃	J4
	178	N	0	1-2	4-CI, 6-F	C(CH ₃) ₂ OCO ₂ CH ₃	J4
50	179	N	0	1-2	4-Cl, 6-F	C(CH ₃) ₂ CH ₂ OH	J4

.- 33 -

	180	N	0	1-2	4-CI, 6-F	C(CH ₃) ₂ CH ₂ OCH ₃	J4
	181	N	0	1-2	4-CI, 6-F	C ₂ H ₅	J4
	182	N	0	1-2	4-CI, 6-F	CO₂Na	J4
	183	Ν	. 0	1-2	4-Cl, 6-F	CONHSO₂CH₃	J4
5	184	Ν	· 0	1-2	4-CI, 6-F	OCH ₂ CO ₂ CH ₃	J4
	185	N	0	1-2	4-CI, 6-F	OCH(CH ₃)CO ₂ CH ₃	J4
	186	N	0	1-2	4-CI, 6-F	OCH ₂ CH=CH ₂	J4
	187	N	0	1-2	4-CI, 6-F	OCH₂C ≡CH	J4
	188	N	0	1-2	4-CI, 6-F	ОН	J4
10	189	N	0	1-2	4-CI, 6-F	OCH₃	J4
	190	Ν	0	1-2	4-Cl, 6-F	OCH(CH ₃) ₂	J4
	191	N	0	1-2	4-CI, 6-F	CH₃	J5
	192	N	0	1-2	4-CI, 6-F	n-C₃H ₇	J5
	193	N	0	1-2	4-CI, 6-F	i-C₃H ₇	J5
1 5	194	N	0	1-2	4-CI, 6-F	t-C₄H ₉	J5
	195	N	0	1-2	4-Cl, 6-F	CH₂OH	J5
	196	Ν	0	1-2	4-CI, 6-F	CH₂CH₂OH	J5
	197	N	0	1-2	4-CI, 6-F	C(CH₃)₂OH	J5
	198	N	0	1-2	4-CI, 6-F	CONHCH₃	J 5
20	199	Ν	0	1-2	4-CI, 6-F	CON(CH ₃) ₂	J5
	200	N	0	1-2	4-CI, 6-F	CO ₂ CH ₃	J 5
	201	N	0	1-2	4-CI, 6-F	Phenyl	J5
	202	Ν	0	1-2	4-CI, 6-F	SCH₃	J5
	2 03	Ν	0	1-2	4-Cl, 6-F	CH₂OCH₃	J5
2 5	204	Ν	0	1-2	4-CI, 6-F	Benzyl	J 5
	205	Ν	0	1-2	4-CI, 6-F	4-chlorophenylmethyl	J5
	206	Ν	0	1-2	4-CI, 6-F	SO ₂ CH ₃	J5
	207	N	0	1-2	4-CI, 6-F	CF₃	J5
	208	N	0	1-2	4-CI, 6-F	$C(CH_3)_2OCO_2CH_3$	J5
30	20 9	N	0	1-2	4-Cl, 6-F	C(CH ₃) ₂ CH ₂ OH	J5
	210	N	0	1-2	4-CI, 6-F	C(CH ₃) ₂ CH ₂ OCH ₃	J 5
	211	Ν	0	1-2	4-CI, 6-F	C_2H_5	J 5
	212	N	0	1-2	4-CI, 6-F	CO₂Na	J5
	213	Ν	О	1-2	4-CI, 6-F	CONHSO₂CH₃	J 5
3 5	214	N	О	1-2	4-CI, 6-F	OCH ₂ CO ₂ CH ₃	J5
	215	N	Ο	1-2	4-CI, 6-F	OCH(CH ₃)CO ₂ CH ₃	J5
	216	Ν	0	1-2	4-CI, 6-F	OCH ₂ CH=CH ₂	J5
	217	Ν	Ο	1-2	4-Cl, 6-F	OCH₂CCH	J5
	218	N	О	1-2	4-CI, 6-F	ОН	J5
4 0	219	N	0	1-2	4-CI, 6-F	OCH ₃	J5
	220	Ν	0	1-2	4-CI, 6-F	OCH(CH ₃) ₂	J5
	221	0	CH	2-3	4-Cl	CH ₃	J1
	222	0	CH	2 -3	4-CI, 6-F	CH ₃	J1
	223	0	CH	2-3	4-CI, 6-F	n-propyl	J1
4 5	224	0	CH	2 -3	4-CI, 6-F	isopropyl	J1
	225	0	CH	2-3	4-CI	n-butyl	J1
	226	0	CH	2-3	4-CI	t-butyl	J1
	227	0	СН	2 -3	4-CI, 6-F	t-butyl	J1
	228	0	CH	2-3	4,6-F ₂	t-butyl	J1
50	229	0	CH	2- 3	4-CI, 6-F	CH(CH ₃)C ₃ H ₇	J1

- 34 -

	230	0	СН	2 -3	4-Cl, 6-F	CH=CH ₂	J1
	231	0	СН	2-3	4-Cl, 6-F	C(CH ₃)=CH ₂	J1
	232	0	СН	2-3	4-CI	CH₂Br	J1
	233	0	. CH	2-3	4-Cl	CHBr ₂	J1
5	234	0	- CH	2-3	4-CI, 6-F	CH(CI)CH₃	J1
	235	Ō	СН	2-3	4-Cl, 6-F	CH(F)CH₃	J1
	236	0	СН	2-3	4-CI, 6-F	CH ₂ CH ₂ CI	J1
	237	0	СН	2-3	4-CI, 6-F	CH₂CH₂F	J1
	238	0	СН	2-3	4-CI	CH₂OH	J1
10	239	0	СН	2-3	4-CI, 6-F	CH₂CH₂OH	J1
	240	0	СН	2-3	4-Cl, 6-F	CH(CH₃)OH	J1
	241	0	СН	2-3	4-CI	C(CH₃)₂OH	J1
	242	O	СН	2-3	4-CI, 6-F	C(CH ₃) ₂ OH	J1
	243	0	СН	2-3	4-CI, 6-F	CH₂CH(CH₃)OH	J1
1 5	244	Ō	СН	2-3	4-CI, 6-F	CH(CH ₃)OC(CH ₃) ₃	J1
	245	Ō	СН	2-3	4-Cl, 6-F	$CH(OC_2H_5)_2$	J1
	246	0	СН	2-3	4-Cl, 6-F	CH(CH ₃)OCOCH ₃	J1
	247	O	СН	2- 3	4-CI, 6-F	CH(CH ₃)OCOCH(CH ₃) ₂	J1
	248	0	СН	2-3	4-Cl, 6-F	CH(CH₃)OCOPh	J1
20	249	0	СН	2-3	4-CI, 6-F	CH(CH ₃)OCONHCH ₃	J1
	250	0	СН	2-3	4-CI, 6-F	CH(CH ₃)OCONHCH ₂ Ph	J1
	251	0	СН	2-3	4-CI	$C(CH_3)_2OCH_3$	J1
	2 52	0	CH	2-3	4-CI, 6-F	$C(CH_3)_2OCH_2OCH_3$	J1
	253	0	СН	2-3	4-Cl, 6-F	C(CH ₃) ₂ OCOCH ₃	J1
2 5	254	0	СН	2-3	4-CI, 6-F	C(CH ₃) ₂ NH ₂	J1
	255	0	СН	2-3	4-CI, 6-F	C(CH ₃) ₂ NHSO ₂ CH ₃	J1
	256	0	СН	2-3	4-CI, 6-F	CH ₂ CH ₂ CH ₂ CN	J1
	257	0	СН	2-3	4-CI	$CH_2N(C_2H_5)_2$	J1
	25 8	0	СН	2-3	4-CI	CH=NOH	J1
30	2 59	0	CH	2-3	4-C1	CH=NOCH ₃	J1
	260	0	СН	2-3	4-Cl, 6-F	CH ₂ CH ₂ OCOCH ₃	J1
	261	0	CH	2-3	4-CI, 6-F	CH2CH2OCONHCH3	J1
	2 62	0	СН	2-3	4-CI, 6-F	CH ₂ CH ₂ CO ₂ H	J1
	2 63	0	СН	2-3	4-CI, 6-F	CH ₂ CH ₂ CO ₂ CH ₃	J1
3 5	264	0	СН	2-3	4-CI	Phenyl	J1
	265	0	CH	2-3	4-CI	CHO	J1
	266	0	СН	2-3	4-CI	CO₂H	J1
	267	0	CH	2-3	Н	$CO_2C_2H_5$	J1
	268	0	СН	2-3	4-CI	$CO_2C_2H_5$	J1
40	269	0	СН	2-3	4-CI	CONH ₂	J1
	270	0	СН	2-3	4-CI	CONHCH₃	J1
	271	0	СН	2-3	4-CI	CON(CH ₃) ₂	J1
	272	0	СН	2-3	4-CI	$NHCO_2C(CH_3)_3$	J1
	273	0	СН	2-3	4-Cl, 6-F	CONH ₂	J1
45	274	0	СН	2-3	4-CI, 6-F	CONH(CH ₃)	J1
	275	0	СН	2-3	4-CI, 6-F	CON(CH₃)₂	J1
	276	0	СН	2 -3	4-CI, 6-F	CO₂H	J1
	277	0	СН	2- 3	4-CI, 6-F	CO ₂ CH ₃	J1
	278	0	СН	2-3	4-CI, 6-F	CH₂OH	J1
50	279	0	СН	2-3	4-CI, 6-F	3,4-dimethoxyphenyl	J1

. - 35 -

	280	0	СН	2-3	4-CI, 6-F	Phenyl	J1
	281	0	СН	2-3	4-CI, 6-F	CH₃	J2
	282	0	СН	2-3	4-CI, 6-F	n-propyl	J2
	283	0	. CH	2 -3	4-Cl, 6-F	isopropyl	J2
5	284	0	CH	2 -3	4-CI, 6-F	t-butyl	J2
	28 5	0	СН	2-3	4-CI, 6-F	CH(CH ₃)C ₃ H ₇	J2
	28 6	0	СН	2-3	4-CI, 6-F	CH=CH₂	J2
	287	0	СН	2-3	4-CI, 6-F	$C(CH_3)=CH_2$	J2
	288	0	СН	2-3	4-CI, 6-F	CH(CI)CH ₃	J2
10	289	0	СН	2-3	4-CI, 6-F	CH(F)CH₃	J2
	290	0	СН	2-3	4-CI, 6-F	CH₂CH₂CI	J2
	291	0	СН	2-3	4-CI, 6-F	CH₂CH₂F	J2
	292	0	СН	2-3	4-CI, 6-F	CH₂CH₂OH	J2
	293	0	СН	2-3	4-CI, 6-F	CH(CH₃)OH	J2
1 5	294	0	СН	2-3	4-CI, 6-F	C(CH ₃) ₂ OH	J2
	295	0	СН	2- 3	4-CI, 6-F	CH₂CH(CH₃)OH	J2
	2 96	0	СН	2-3	4-CI, 6-F	$CH(CH_3)OC(CH_3)_3$	J2
	2 97	0	СН	2-3	4-CI, 6-F	$CH(OC_2H_5)_2$	J2
	2 98	0	СН	2-3	4-CI, 6-F	CH(CH ₃)OCOCH ₃	J2
20	29 9	0	CH	2-3	4-CI, 6-F	$CH(CH_3)OCOCH(CH_3)_2$	J2
	30 0	0	СН	2 -3	4-CI, 6-F	CH(CH₃)OCOPh	J2
	301	0	СН	2-3	4-CI, 6-F	CH(CH₃)OCONHCH₃	J2
	302	0	СН	2-3	4-Cl, 6-F	CH(CH ₃)OCONHCH ₂ Ph	J2
	30 3	0	СН	2-3	4-CI, 6-F	$C(CH_3)_2OCH_2OCH_3$	J2
25	304	0	ĊН	2 -3	4-CI, 6-F	C(CH ₃) ₂ OCOCH ₃	J2
	30 5	0	СН	2 -3	4-Cl, 6-F	$C(CH_3)_2NH_2$	J2
	306	0	СН	2 -3	4-CI, 6-F	C(CH ₃) ₂ NHSO ₂ CH ₃	J2
	307	0	СН	2-3	4-CI, 6-F	CH ₂ CH ₂ CH ₂ CN	J2
	308	0	СН	2-3	4-CI, 6-F		J2
30	30 9	0	СН	2-3	4-CI, 6-F	CH2CH2OCONHCH3	J2
	310	0	СН	2 -3	4-CI, 6-F	CH ₂ CH ₂ CO ₂ H	J2
	311	0	СН	2-3	4-CI, 6-F	CH ₂ CH ₂ CO ₂ CH ₃	J2
	312	0	СН	2-3	4-CI, 6-F	CONH₂	J2
	31 3	0	СН	2-3	4-CI, 6-F	CONH(CH ₃)	J2
35	314	0	СН	2 -3	4-CI, 6-F	CON(CH ₃) ₂	J2
	3 15	0	СН	2 -3	4-CI, 6-F	CO₂H	J2
	3 16	0	СН	2 -3	4-CI, 6-F	CO ₂ CH ₃	J2
	317	0	СН	2-3	4-CI, 6-F	CH₂OH	J2
	3 18	0	СН	2-3	4-CI, 6-F	3,4-dimethoxyphenyl	J2
40	319	0	СН	2- 3	4-CI, 6-F	Phenyl	J2
	320	0	СН	2-3	4-CI, 6-F	CH₃	J3
	321	0	СН	2-3	4-CI, 6-F	C ₂ H ₅	J3
	322	0	СН	2-3	4-CI, 6-F	CH(CI)CH₃	J3
	32 3	0	СН	2-3	4-C1, 6-F	CH(F)CH₃	J3
4 5	324	0	CH	2-3	4-CI, 6-F	CH ₂ CH ₂ Cl	J3
	325	0	CH	2-3	4-CI, 6-F	CH₂CH₂F	J3
	326	0	CH	2-3	4-Cl, 6-F	CH ₂ CH ₂ OH	J3
	327	0	СН	2-3	4-CI, 6-F	CH(CH₃)OH	J3
	3 28	0	СН	2-3	4-Cl, 6-F	C(CH₃)₂OH	J3
50	329	0	СН	2-3	4-CI, 6-F	C(CH3)2OCH2OCH3	J3

.- 36 -

	000	•	СН	2-3	4-Cl, 6-F	C(CH3)2NHSO2CH3	J3
	330 331	0	CH	2-3	4-Cl, 6-F	CH ₂ CH ₂ CH ₂ CN	J3
	332	0	СН	2-3	4-CI, 6-F	CH2CH2CO2CH3	J3
	333	0	- CH	2-3	4-CI, 6-F	CON(CH ₃) ₂	J3
5	334	0	СН	2-3	4-Cl, 6-F	CH₃	J4
	33 5	0	СН	2-3	4-CI, 6-F	C ₂ H ₅	J4 J4
	33 6	0	СН	2-3	4-Cl, 6-F	CH(CI)CH₃	J4 J4
	337	0	CH	2-3	4-Cl, 6-F	CH(F) CH ₃ CH₂CH₂Cl	J4
	338	0	CH	2-3	4-Cl, 6-F 4-Cl, 6-F	CH ₂ CH ₂ F	J4
10	339	0	CH	2-3 2-3	4-Cl, 6-F	CH ₂ CH ₂ OH	J4
	340	0	CH CH	2-3 2-3	4-Cl, 6-F	CH(CH ₃)OH	J4
	341	0	CH	2-3 2-3	4-Cl, 6-F	C(CH ₃) ₂ OH	J4
	342	_		2-3	4-Cl, 6-F	C(CH ₃) ₂ OCH ₂ OCH ₃	J4
	343	0	CH	2-3 2-3	4-Cl, 6-F	C(CH ₃) ₂ NHSO ₂ CH ₃	J4
15	344 345	0	CH CH	2-3 2-3	4-Cl, 6-F	CH ₂ CH ₂ CH ₂ CN	J4
	346	0	СН	2 -3	4-Cl, 6-F	CH2CH2CO2CH3	J4
	347	0	CH	2 -3	4-CI, 6-F	CON(CH ₃) ₂	J4
	348	0	CH	2-3	4-Cl, 6-F	CH₃	J 5
20	349	0	CH	2-3	4-CI, 6-F	C ₂ H ₅	J5
20	350	0	СН	2-3	4-CI, 6-F	CH(CI)CH₃	J 5
	351	o	СН	2-3	4-CI, 6-F	CH(F)CH₃	J5
	352	0	СН	2-3	4-CI, 6-F	CH₂CH₂CI	J5
	35 3	0	СН	2- 3	4-CI, 6-F	CH ₂ CH ₂ F	J5
2 5	354	0	CH	2 -3	4-CI, 6-F	CH₂CH₂OH	J5
	35 5	0	СН	2-3	4-CI, 6-F	CH(CH₃)OH	J5 J5
	356	0	CH	2-3	4-CI, 6-F	C(CH ₃) ₂ OH	J5 J5
	357	0	СН	2-3	4-CI, 6-F	C(CH ₃) ₂ OCH ₂ OCH ₃	J5
	35 8	0	CH	2 -3	4-Cl, 6-F	C(CH3)2NHSO2CH3 CH2CH2CH2CN	J5
30	35 9	0	СН	2-3	4-Cl, 6-F	CH ₂ CH ₂ CO ₂ CH ₃	J5
	3 60	0	CH	2-3	4-CI, 6-F		J5
	361	0	CH N	2-3 2- 3	4-CI, 6-F 4-CI, 6-F	H	J1
	362	NH	N	2 -3	4-Cl, 6-F	CH ₃	J1
3 5	3 63 3 64	NH NH	N	2-3	4-Cl, 6-F	CHF ₂	J1
35	36 5	NH	N N	2-3	4-Cl, 6-F	CF ₃	J1
	36 6	NH	N N	2-3	4-CI, 6-F	CCIF ₂	J1
	3 67	NH	N	2 -3	4-Cl, 6-F	C ₂ H ₅	J1
	36 8	NH	N	2-3	4-CI, 6-F	i-C₃H ₇	J1
4 0	36 9	NH	N	2 -3	4-CI, 6-F	t-C₄H ₉	J1
	370	NH	N	2-3	4-CI, 6-F	CH₂OCH₃	J1
	371	NH	N	2-3	4-CI, 6-F	$C(CH_3)_2OC(O)CH_3$	J1
	3 72	NH	N	2-3	4-CI, 6-F	$C_2H_4CO_2C_2H_5$	J1
	3 73	NH	N	2- 3	4-CI, 6-F	Cyclohexyl	J1
4 5	374	NH	N	2-3	4-Cl, 6-F	Adamantyl	J1
	3 75	NH	N	2-3	4-CI, 6-F	Phenyl	J1
	3 76	NH	N	2-3	4-CI, 6-F	Benzyl	J1
	377	NH	N	2-3	4-Cl, 6-F	CH(CH ₃)C ₆ H ₅	J1 J1
	3 78	NH	N	2 -3	4-CI, 6-F	CH ₂ OC ₈ H ₅	J1 J1
50	37 9	NH	N	2-3	4-CI, 6-F	$C_2H_4C_6H_5$	J 1

- 37 -

	38 0	NH	N	2-3	4-CI, 6-F	$C_3H_6C_6H_5$	J1
	381	NH	N	2-3	4-CI, 6-F	2-chlorophenylmethyl	J1
	382	NH	N	2-3	4-CI, 6-F	3-chlorophenylmethyl	J1
	38 3	NH -	N	2-3	4-Cl, 6-F	4-chlorophenylmethyl	J1
5	384	NH	N	2-3	4-CI, 6-F	CF ₂ CF ₃	J1
	38 5	NH	N	2-3	4-CI, 6-F	Furan-2-yl	J1
	38 6	NH	N	2-3	4-Cl, 6-F	CH₂CI	J1
	387	NH	N	2-3	4-Cl, 6-F	C(CH ₃) ₂ CH ₂ Cl	J1
	38 8	NH	N	2-3	4-CI, 6-F	OC₂H₅	J1
10	389	N	NH	1-2	4-CI, 6-F	CH₃	J1
	390	N	NH	1-2	4-CI, 6-F	C₂H₅	J1
	391	N	NH	1-2	4-CI, 6-F	isop ropy l	J1
	392	N	NH	1-2	4-Cl, 6-F	t-butyl	J1
	39 3	N	NH	1-2	4-Cl, 6-F	CF ₃	J1
1 5	394	N	NH	1-2	4-Cl, 6-F	CF₂CF₃	J1
	395	N	NCH₃	1-2	4-Cl, 6-F	CH ₃	J1
	39 6	N	NCH₃	1-2	4-Cl, 6-F	C₂H₅	J1
	397	N	NCH ₃	1-2	4-CI, 6-F	isopr opy l	J1
	398	N	NCH ₃	1-2	4-CI, 6-F	t-butyl	J1
20	39 9	N	NCH ₃	1-2	4-Cl, 6-F	CF ₃	J1
	400	N	NCH ₃	1-2	4-Cl, 6-F	CF₂CF₃	J1
	401	N	NCH ₃	1-2	4-CI, 6-F	CO ₂ CH ₂ CH ₃	J1
	402	N	NC ₂ H ₅	1-2	4-CI, 6-F	CH₃	J1
	403	N	NC ₂ H ₅	1-2	4-CI, 6-F	C₂H₅	J1
2 5	4 04	NH	NH		4-NO₂, 6-F	CF₃	J1
	40 5	N ⁻ H₃N ⁺ CH(CH₃)₂	N	2-3	4-CI, 6-F	CH ₃	J1
	40 6	NCH ₃	N	2-3	4-CI, 6-F	CF ₃	J1
	407	NCH ₃	NC ₂ H ₅	1-2	4-Cl, 6-F	isopropyl	J1
	408	N	NC ₂ H ₅	1-2	4-CI, 6-F	t-butyl	J1
30	409	N	NC ₂ H ₅	1-2	4-Cl, 6-F	CF ₃	J1
30	410	N	NC ₂ H ₅	1-2	4-CI, 6-F	CF ₂ CF ₃	J1
	411	N	NC₄H ₉	1-2	4-CI, 6-F	CH ₃	J1
	412	N	NC₄H ₉	1-2	4-Cl, 6-F	C_2H_5	J1
	413	N	NC₄H ₉	1-2	4-CI, 6-F	isopropyl	J1
35	414	N	NC₄H ₉	1-2	4-CI, 6-F	t-butyl	J1
	415	N	NC₄H ₉	1-2	4-CI, 6-F	CF₃	J1
	416	N	NC₄H ₉	1-2	4-CI, 6-F	CF ₂ CF ₃	J1
	417	N	NCH ₂ OCH ₃	1-2	4-Cl, 6-F	CH₃	J1
	418	N	NCH ₂ OCH ₃	1-2	4-CI, 6-F	C ₂ H ₅	J1
40	419	. N	NCH ₂ 0CH ₃	1-2	4-CI, 6-F	isopropyl	J1
10	420	N	NCH ₂ 0CH ₃	1-2	4-CI, 6-F	t-butyl	J1
	421	N	NCH ₂ OCH ₃	1-2	4-CI, 6-F	CF ₃	J1
	422	N	NCH ₂ OCH ₃	1-2	4-CI, 6-F	CF ₂ CF ₃	J1
	423	N	NCO ₂ CH ₃	1-2	4-CI, 6-F	CH₃	J1
45	424	N	NCO ₂ CH ₃	1-2	4-Cl, 6-F	C ₂ H ₅	J1
	425	N N	NCO ₂ CH ₃	1-2	4-CI, 6-F	isopropyl	J1
	426	N	NCO ₂ CH ₃	1-2	4-Cl, 6-F	t-butyl	J1
	427	N	NCO ₂ CH ₃	1-2	4-CI, 6-F	CF ₃	J1
	428	N	NCO ₂ CH ₃	1-2	4-Cl, 6-F	CF ₂ CF ₃	J1
	5	• •	2 - · · 3		•	-	

	429	N		NSO₂CH₃	1-2	4-CI, 6-F	CH₃	J1	
	430	N		NSO₂CH₃	1-2	4-Cl, 6-F	C ₂ H ₅	⁻ J1	
	431	N		NSO ₂ CH ₃	1-2	4-Cl, 6-F	isopropyl	J1	
	432	N	-	NSO₂CH₃	1-2	4-Cl, 6-F	t-butyl	J1	
5	433	N	-	NSO ₂ CH ₃	1-2	4-CI, 6-F	CF ₃	J1	
	434	N		NSO ₂ CH ₃	1-2	4-Cl, 6-F	CF ₂ CF ₃	J1	
	435	N		NCH₂CHCH₂	1-2	4-CI, 6-F	CH₃	J1	
	436	N		NCH ₂ CHCH ₂	1-2	4-CI, 6-F	C₂H₅	J1	
	437	N		NCH ₂ CHCH ₂	1-2	4-CI, 6-F	isop ropy l	J1	
10	438	N		NCH ₂ CHCH ₂	1-2	4-Cl, 6-F	t-butyl	J1	
	439	N		NCH ₂ CHCH ₂	1-2	4-CI, 6-F	CF ₃	J1	
	440	N		NCH ₂ CHCH ₂	1-2	4-CI, 6-F	CF ₂ CF ₃	J1	
	441	N		NCH₂CCH	1-2	4-Cl, 6-F	CH₃	J1	
	442	N		NCH₂CCH	1-2	4-Cl, 6-F	C₂H₅	J 1	
1 5	443	N		NCH₂CCH	1-2	4-CI, 6-F	isop ropy l	J 1	
	444	N		NCH₂CCH	1-2	4-CI, 6-F	t-butyl	J1	
	445	N		NCH₂CCH	1-2	4-Cl, 6-F	CF₃	J1	
	446	N		NCH₂CCH	1-2	4-CI, 6-F	CF₂CF₃	J1	
	447	N		NCH₂CO₂Me	1-2	4-CI, 6-F	CH₃	J1	
20	448	N		NCH₂CO₂Me	1-2	4-Cl, 6-F	C₂H₅	J1	
	449	N		NCH ₂ CO ₂ Me	1-2	4-CI, 6-F	isopr opy l	J1	
	450	N		NCH ₂ CO ₂ Me	1-2	4-CI, 6-F	t-butyl	J1	
	451	N		NCH₂CO₂Me	1-2	4-Cl, 6-F	CF₃	J1	
	452	N		NCH₂CO₂Me	1-2	4-Cl, 6-F	CF ₂ CF ₃	J1	
2 5	453	N		NCF ₃	1-2	4-CI, 6-F	CH ₃	J1	
	454	N		NCF ₃	1-2	4-CI, 6-F	C ₂ H ₅	J1	
	455	N		NCH₂CO₂Me	1-2	4-Cl, 6-F	isopropyl	J1	
	456	N		NCH₂CO₂Me	1-2	4-Cl, 6-F	t-butyl	J1	
	457	N		NCH₂CO₂ M e	1-2	4-Cl, 6-F	CF₃	J1	
30	458	N		NCF ₃	1-2	4-CI, 6-F	CF ₂ CF ₃	J1	
	45 9	NH		N	2-3	4-CI, 6-F	CH₃	J2	
	4 60	NH		N	2-3	4-Cl, 6-F	C ₂ H ₅	J2 J2	
	4 61	NH		N	2 -3	4-Cl, 6-F	isopropyl	J2 J2	
	462	NH		N	2-3	4-Cl, 6-F	t-butyl	J2 J2	
3 5	4 63	NH		N	2-3	4-Cl, 6-F	CF₃	J2 J2	
	4 64	NH		N	2-3	4-Cl, 6-F	CF₂CF₃ CH₃	J3	
	46 5	NH		N	2-3	4-Cl, 6-F	C₂H₅	J3	
	466	NH		N	2-3	4-Cl, 6-F	isopropyl	J3	
	467	NH		N	2-3	4-Cl, 6-F 4-Cl, 6-F	t-butyl	J3	
40	468	NH		N	2-3 2-3	4-Cl, 6-F	CF ₃	J3	
	469	NH		N .	2-3 2-3	4-Cl, 6-F	CF ₂ CF ₃	J3	
	470	NH		N	2-3 2-3	4-Cl, 6-F	CH ₃	J4	
	471	NH		N N	2-3 2-3	4-Cl, 6-F	C₂H₅	J4	
4 =	472	NH		N N	2-3 2-3	4-CI, 6-F	isopropyl	J4	
45	473	NH			2-3 2-3	4-Cl, 6-F	t-butyl	J4	
	474 475	NH		N	2-3 2-3	4-Cl, 6-F	CF ₃	J4	
	475	NH		N N	2-3 2-3	4-Cl, 6-F	CF ₂ CF ₃	J4	
	476 477	NH NH		N	2-3 2-3	4-Cl, 6-F	CH ₃	J 5	
50	477 478	NH NH		N	2-3 2-3	4-Cl, 6-F	C₂H₅	J5	
50	4/0	INIT		• •		,	2 3		

	470	MLI	N	2-3	4-CI, 6-F	isopropyl	J5
	479	NH	N	2-3	4-Cl, 6-F	t-butyl	J5
	480	NH NH	N	2-3	4-Cl, 6-F	CF ₃	J5
	481 482	NH	. N	2-3	4-Cl, 6-F	CF ₂ CF ₃	J5
5		NH	NH	1-2	4-Cl, 6-F	ĆH₃ °	J1
5	483 484	CH	NH	1-2	4-Cl, 6-F	n-C ₃ H ₇	J1
	485	CH	NH	1-2	4-CI, 6-F	i-C₃H ₇	J1
	48 6	CH	. NH	1-2	4-Cl, 6-F	t-C ₄ H ₉	J1
	487	СН	NH	1-2	4-Cl, 6-F	CH₂OH	J1
3.0	488	СН	NH	1-2	4-CI, 6-F	CH ₂ CH ₂ OH	J1
10	489	CH	NH	1-2	4-Cl, 6-F	C(CH ₃) ₂ OH	J1
	49 9	СН	NH	1-2	4-Cl, 6-F	CONHCH ₃	J1
	49 0 49 1	CH	NH	1-2	4-Cl, 6-F	CON(CH ₃) ₂	J1
	491 492	СН	NH	1-2	4-Cl, 6-F	CO ₂ CH ₃	J1
15	49 2 49 3	CH	NH	1-2	4-Cl, 6-F	CO ₂ CH ₂ CH ₃	J1
1 5	49 3 49 4	CH	NH	1-2	4-CI, 6-F	Phenyl	J1
	495 495	СН	NH	1-2	4-Cl, 6-F	CF ₂ CF ₃	J1
	49 5	CH	NH	1-2	4-Cl, 6-F	CH₂OCH₃	J1
	49 0 49 7	CH	NH	1-2	4-Cl, 6-F	Benzyl	J1
20	49 8	СН	NH	1-2	4-CI, 6-F	4-chlorophenylmethyl	J1
20	49 9	CH	NH	1-2	4-Cl, 6-F	SO₂CH₃	J1
	50 0	CH	NH	1-2	4-Cl, 6-F	CF ₃	J1
	5 01	CH	NH	1-2	4-Cl, 6-F	C(CH ₃) ₂ OCOCH ₃	J1
	502	CH	NH	1-2	4-Cl, 6-F	C(CH ₃) ₂ CH ₂ OH	J1
2 5	50 2	CH	NH	1-2	4-CI, 6-F	C(CH ₃) ₂ CH ₂ OCH ₃	J1
25	5 03	CH	NH	1-2	4-Cl, 6-F	C ₂ H ₅	J1
	504 505	CH	NH	1-2	4-CI, 6-F	CO ₂ Na	J1
	50 6	CH	NH	1-2	4-Cl, 6-F	CONHSO ₂ CH ₃	J1
	50 7	CH	NH	1-2	4-Cl, 6-F	CHFCH ₃	J1
30	50 7	CH	NH	1-2	4-CI, 6-F	CH ₂ CO ₂ CH ₂ CH ₃	J1
30	50 9	CH	NCH₃	1-2	4-Cl, 6-F	CH ₃	J1
	510	CH	NCH ₃	1-2	4-CI, 6-F	C ₂ H ₅	J1
	511	CH	NCH ₃	1-2	4-Cl, 6-F	isopropyl	J1
	5 12	CH	NCH ₃	1-2	4-CI, 6-F	t-butyl	J1
35	513	CH	NCH ₃	1-2	4-CI, 6-F	CF ₃	J1
33	514	СН	NCH ₃	1-2	4-CI, 6-F	CF ₂ CF ₃	J1
	5 15	CH	NCH ₃	1-2	4-CI, 6-F	CHFCH₃	J1
	5 16	CH	NCH ₃	1-2	4-CI, 6-F	CON(CH ₃) ₂	J1
	517	CH	NCH ₃	1-2	4-CI, 6-F	CH ₂ CO ₂ C ₂ H ₅	J1
40	518	СН	NCH ₃	1-2	4-CI, 6-F	CH ₂ CH ₂ CN	J1
40	5 19	СН	NCH ₃	1-2	4-Cl, 6-F	C(CH ₃) ₂ OH	J1
	520	СН	NCH ₃	1-2	4-CI, 6-F	C(CH ₃) ₂ OCOCH ₃	J1
	521	CH	NCH ₃	1-2	4-CI, 6-F	C(CH ₃) ₂ NHSO ₂ CH ₃	J1
	522	СН	NCH ₃	1-2	4-Cl, 6-F	CO ₂ CH ₂ CH ₃	J1
4 5	52 3	СН	NC ₂ H ₅	1-2	4-CI, 6-F	CH ₃	J1
40	524	СН	NC ₂ H ₅	1-2	4-Cl, 6-F	C ₂ H ₅	J1
	525	CH	NC ₂ H ₅	1-2	4-Cl, 6-F	isopropyl	J1
	52 6	CH	NC ₂ H ₅	1-2	4-CI, 6-F	t-butyl	J1
	527	CH	NC ₂ H ₅	1-2	4-CI, 6-F	CF ₃	J1
50	528	CH	NC ₂ H ₅	1-2	4-Cl, 6-F	CO ₂ CH ₃	J1
50	520	J. 1			-	· · · · · · · · · · · · · · · · · · ·	

- 40 -

	52 9	СН		NC₄H ₉	1-2	4-Cl, 6-F	CH ₃	J1
	530	СН		NC₄H ₉	1-2	4-CI, 6-F	C₂H₅	- J1
	531	СН		NC₄H ₉	1-2	4-CI, 6-F	isopropyl	J1
	5 32	CH		NC₄H ₉	1-2	4-CI, 6-F	t-butyl	J1
5	53 3	СН	´-	NC₄H ₉	1-2	4-Cl, 6-F	CF ₃	J1
	5 34	CH		NC₄H ₉	1-2	4-CI, 6-F	CO₂CH₃	J1
	53 5	СН		NCH ₂ OCH ₃	1-2	4-CI, 6-F	CH₃	J1
	53 6	СН		NCH ₂ OCH ₃	1-2	4-CI, 6-F	C₂H₅	J1
	5 37	CH		NCO ₂ CH ₃	1-2	4-CI, 6-F	isop ropyl	J1 Ţ
10	53 8	СН		NCH ₂ OCH ₃	1-2	4-CI, 6-F	t-butyl	J1
	53 9	СН		NCH ₂ OCH ₃	1-2	4-CI, 6-F	CF ₃	J1
	54 0	СН		NCH ₂ OCH ₃	1-2	4-CI, 6-F	CO ₂ CH ₃	J1
	541	СН		NCO ₂ CH ₃	1-2	4-CI, 6-F	CH₃	J1
	542	СН		NCO ₂ CH ₃	1-2	4-CI, 6-F	C₂H₅	J1
15	54 3	СН		NCO ₂ CH ₃	1-2	4-CI, 6-F	isop ropy l	J1
	544	СН		NCO ₂ CH ₃	1-2	4-CI, 6-F	t-butyl	J1
	54 5	СН		NCO ₂ CH ₃	1-2	4-CI, 6-F	CF ₃	J1
	5 46	СН		NCO ₂ CH ₃	1-2	4-CI, 6-F	CO ₂ CH ₃	J1
	547	СН		NSO ₂ CH ₃	1-2	4-Cl, 6-F	CH₃	J1
20	5 48	СН		NSO ₂ CH ₃	1-2	4-CI, 6-F	C₂H₅	J1
	54 9	СН		NSO ₂ CH ₃	1-2	4-CI, 6-F	isop ropy l	J1
	55 0	СН		NSO ₂ CH ₃	1-2	4-CI, 6-F	t-butyl	J1
	5 51	СН		NSO ₂ CH ₃	1-2	4-Cl, 6-F	CF ₃	J1
	5 52	СН		NSO ₂ CH ₃	1-2	4-CI, 6-F	CO ₂ CH ₃	J1
25	55 3	СН		NCH ₂ CHCH ₂	1-2	4-CI, 6-F	CH ₃	J1
	5 54	СН		NCH ₂ CHCH ₂	1-2	4-Cl, 6-F	C ₂ H ₅	J1
	55 5	СН		NCH ₂ CHCH ₂	1-2	4-CI, 6-F	isopropyl	J1
	5 56	СН		NCH ₂ CHCH ₂	1-2	4-Cl, 6-F	t-butyl	J1
	5 57	СН		NCH ₂ CHCH ₂	1-2	4-CI, 6-F	CF ₃	J1
30	55 8	СН		NCH ₂ CHCH ₂	1-2	4-Cl, 6-F	CO ₂ CH ₃	J1
	5 59	СН		NCH₂C≣CH	1-2	4-CI, 6-F	CH ₃	J1
	56 0	СН		NCH ₂ C≡CH	1-2	4-CI, 6-F	C ₂ H ₅	J1
	5 61	СН		NCH₂C≣CH	1-2	4-Cl, 6-F	isopropyl	J1
	56 2	СН		NCH₂C≣CH	1-2	4-CI, 6-F	t-butyl	J1
3 5	5 63	СН		NCH ₂ C≡CH	1-2	4-CI, 6-F	CF ₃	J1
	5 64	СН		NCH ₂ C≣CH	1-2	4-CI, 6-F	CO₂CH₃	J1
	5 65	СН		NCH ₂ CO ₂ Me	1-2	4-CI, 6-F	CH₃	J1
	5 66	СН		NCH ₂ CO ₂ Me	1-2	4-Cl, 6-F	C_2H_5	J1
	5 67	СН		NCH ₂ CO ₂ Me	1-2	4-CI, 6-F	isopropyl	J1
40	5 68	СН		NCH ₂ CO ₂ Me	1-2	4-CI, 6-F	t-butyl	J1
	5 69	СН		NCH ₂ CO ₂ Me	1-2	4-CI, 6-F	CF₃	J1
	57 0	СН		NCH ₂ CO ₂ Me	1-2	4-CI, 6-F	CO₂CH₃	J1
	571	СН		NCH ₂ CHF ₂	1-2	4-CI, 6-F	CH₃	J1
	572	СН		NCH ₂ CHF ₂	1-2	4-Cl, 6-F	C_2H_5	J1
45	5 73	СН		NCH ₂ CHF ₂	1-2	4-Cl, 6-F	isopropyl	J1
	574	СН		NCH ₂ CHF ₂	1-2	4-CI, 6-F	t-butyl	J1
	57 5	СН		NCH ₂ CHF ₂	1-2	4-CI, 6-F	CF ₃	J1
	5 76	СН		NCH ₂ CHF ₂	1-2	4-CI, 6-F	CO ₂ CH ₃	J1
	577	СН		NH	1-2	4-CI, 6-F	CH₃	J2
50	578	СН		NH	1-2	4-CI, 6-F	C_2H_5	J2
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- 41 -

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	5 79	СН	NH	1-2	4-Cl, 6-F	isopropyl	J2
	580	СН	NH	1-2	4-Cl. 6-F	t-butyl	J2
	5 81	СН	NH	1-2	4-CI, 6-F	CF₃	J2
	582	СН	- NH	1-2	4-CI, 6-F	CO₂CH₃	J2
5	58 3	СН	NH	1-2	4-CI, 6-F	CH ₃	J3
	584	СН	NH	1-2	4-Cl, 6-F	C₂H₅	J3
	58 5	СН	NH	1-2	4-Cl, 6-F	isopropyl	J3
	58 6	СН	NH	1-2	4-CI, 6-F	t-butyl	J3
	5 87	СН	NH	1-2	4-CI, 6-F	CF₃	J3
10	58 8	CH	NH	1-2	4-CI, 6-F	CO₂CH₃	J3
	58 9	СН	NH	1-2	4-Cl, 6-F	CH₃	J4
	59 0	СН	NH	1-2	4-CI, 6-F	C₂H₅	J4
	591	CH	NH	1-2	4-CI, 6-F	isopropyl	J4
	59 2	СН	NH	1-2	4-Cl, 6-F	t-butyl	J4
1 5	59 3	CH	NH	1-2	4-CI, 6-F	CF ₃	J4
	594	СН	NH	1-2	4-CI, 6-F	CO₂CH₃	J4 IE
	5 95	CH	NH	1-2	4-Cl	CO ₂ CH ₂ CH ₃	J5
	59 6	CH	NH	1-2	4-Cl, 6-F	CH ₃	J5
	5 97	СН	NH	1-2	4-CI, 6-F	C₂H₃	J5
20	59 8	CH	NH	1-2	4-CI, 6-F	isopropyl	J5
	59 9	СН	NH	1-2	4-CI, 6-F	t-butyl	J5
	60 0	CH	NH	1-2	4-CI, 6-F	CF ₃	J5
	601	CH	NH	1-2	4-CI, 6-F	CO ₂ CH ₃	J5
	602	NH	СН	2 -3	4-CI, 6-F	CH ₃	J7
2 5	60 3	NH	CH	2-3	4-CI, 6-F	n-C ₃ H ₇	J1
	604	NH	СН	2-3	4-CI, 6-F	i-C ₃ H ₇	J1
	60 5	NH	СН	2-3	4-Cl, 6-F	t-C₄H ₉	J1
	60 6	NH	СН	2-3	4-Cl, 6-F	CH₂OH	J1
	60 7	NH	CH	2-3	4-CI, 6-F	CH₂CH₂OH	J1
30	60 8	NH	СН	2-3	4-Cl, 6-F	C(CH₃)₂OH	J1
	60 9	NH	CH	2-3	4-CI, 6-F	CONHCH₃	J1
	61 0	NH	СН	2-3	4-Cl, 6-F	CON(CH ₃) ₂	J1
	611	NH	CH	2-3	4-CI, 6-F	CO₂CH₃	J1
	612	NH	CH	2 -3	4-Cl, 6-F	Phenyl	J1
3 5	61 3	NH	CH	2-3	4-CI, 6-F	CF ₂ CF ₃	J1
	614	NH	СН	2 -3	4-CI, 6-F	CH ₂ OCH ₃	J1
	61 5	NH	CH	2-3	4-CI, 6-F	Benzyl	J1
	6 16	NH	СН	2 -3	4-CI, 6-F	4-chlorophenylmethyl	J1
	617	NH	CH	2-3	4-Cl, 6-F	SO₂CH₃	J1
40	61 8	NH	СН	2-3	4-CI, 6-F	CF ₃	J1
	61 9	NH	CH	2-3	4-CI, 6-F	C(CH ₃) ₂ OCOCH ₃	J1
	62 0	NH	CH	2-3	4-CI, 6-F	C(CH ₃) ₂ CH ₂ OH	J1
	621	NH	CH	2- 3	4-CI, 6-F	C(CH ₃) ₂ CH ₂ OCH ₃	J1
	622	NH	СН	2-3	4-CI, 6-F	C ₂ H ₅	J1
45	62 3	NH	СН	2-3	4-CI, 6-F	CO₂Na	J1
	624	NH	СН	2-3	4-CI, 6-F	CONHSO ₂ CH ₃	J1
	62 5	NH	СН	2-3	4-Cl, 6-F	CHFCH₃	J1
	62 6	NH	СН	2-3	4-CI, 6-F	CH ₂ CO ₂ CH ₂ CH ₃	J1
	627	NH	СН	2-3	4-Cl, 6-F	CH₃	J2
50	62 8	NH	СН	2-3	4-CI, 6-F	C_2H_5	J2

- 42 -

	62 9	NH	СН	2-3	4-CI, 6-F	isop ropy l	J2
	63 0	NH	СН	2-3	4-Cl, 6-F	t-butyl	J2
	631	NH	СН	2-3	4-CI, 6-F	CF ₃	J2
	632	NH	- CH	2-3	4-CI, 6-F	CO₂CH₃	J2
5	63 3	NH	СН	2-3	4-Cl, 6-F	CH₃	J3
	634	NH	СН	2-3	4-CI, 6-F	C₂H₅	J 3
	63 5	NH	СН	2-3	4-Cl, 6-F	isopropyl	J 3
	63 6	NH	СН	2-3	4-CI, 6-F	t-butyl	J3
	6 37	NH	СН	2-3	4-CI, 6-F	CF ₃	J3
10	63 8	NH	СН	2-3	4-CI, 6-F	CO ₂ CH ₃	J3
	63 9	NH	CH	2-3	4-CI, 6-F	CH₃	J4
	64 0	NH	CH	2-3	4-Cl, 6-F	C₂H₅	J4
	641	NH	CH	2-3	4-Cl, 6-F	isopr opyl	J4
	642	NH	CH	2-3	4-CI, 6-F	t-butyl	J4
1 5	643	NH	СН	2-3	4-Cl, 6-F	CF₃	J4
	644	NH	СН	2-3	4-CI, 6-F	CO₂CH₃	J4
	64 5	NH	CH	2-3	4-CI, 6-F	CH₃	J 5
	64 6	NH	СН	2-3	4-Cl, 6-F	C₂H₅	J 5
	647	NH	СН	2 -3	4-CI, 6-F	isopr opyl	J5
20	64 8	NH	СН	2-3	4-CI, 6-F	t-butyl	J5
	64 9	NH	СН	2-3	4-CI, 6-F	CF₃	J5
	65 0	NH	CH	2-3	4-Cl, 6-F	CO₂CH₃	J5
	651	NH	CCH₃	2-3	4-CI, 6-F	CH₃	J1
	652	NH	CCH₃	2-3	4-CI, 6-F	C ₂ H ₅	J1
25	65 3	NH	CCH₃	2 -3	4-CI, 6-F	isop ropyl	J1
	654	NH	CCH₃	2-3	4-CI, 6-F	t-buty!	J1
	65 5	NH	CCH₃	2-3	4-CI, 6-F	CF ₃	J1
	65 6	NH	CCH₃	2-3	4-CI, 6-F	CO ₂ CH ₃	J1
	657	NH	CCH₂CH₃	2-3	4-Cl, 6-F	CH₃	J1
30	65 8	NH	CCH₂CH₃	2 -3	4-Cl, 6-F	C₂H₅	J1
	65 9	NH	CCH₂CH₃	2-3	4-CI, 6-F	isopropyl	J1
	6 60	NH	CCH₂CH₃	2- 3	4-CI, 6-F	t-butyl	J1
	6 61	NH	CCH₂CH₃	2 -3	4-CI, 6-F	CF ₃	J1
	6 62	NH	CCH₂CH₃	2 -3	4-Cl, 6-F	CO ₂ CH ₃	J1
35	6 63	NH	CCH₂CHF₂	2 -3	4-CI, 6-F	CH ₃	J1
	6 64	NH	CCH ₂ CHF ₂	2-3	4-CI, 6-F	C₂H₅	J1
	6 65	NH	CCH₂CHF₂	2 -3	4-Cl, 6-F	isopropyl	J1
	6 66	NH	CCH₂CHF₂	2 -3	4-CI, 6-F	t-butyl	J1
	6 67	NH	CCH ₂ CHF ₂	2 -3	4-CI, 6-F	CF ₃	J1
40	6 68	NH	CCH₂CHF₂	2-3	4-Cl, 6-F	CO ₂ CH ₃	J1
	6 69	NH	СН	2-3	4-Cl, 6-F	CH₃	J2
	6 70	NH	СН	2-3	4-CI, 6-F	C_2H_5	J2
	671	NH	СН	2- 3	4-CI, 6-F	isopropyl	J2
	672	NH	СН	2-3	4-CI, 6-F	t-butyl	J2
45	673	NH	CH	2 -3	4-CI, 6-F	CF₃	J2
	674	NH	СН	2- 3	4-CI, 6-F	CO₂CH₃	J2
	67 5	NH	СН	2 -3	4-CI, 6-F	CH₃	J3
	6 76		СН	2- 3	4-CI, 6-F	C ₂ H ₅	J3
	677		СН	2-3	4-CI, 6-F	isopropyl	J3
50	6 78		СН	2- 3	4-CI, 6-F	t-butyl	J3

- 43 -

			011	0.0	40165	CF ₃	J3
	679	NH	CH	2-3	4-Cl, 6-F 4-Cl, 6-F	CO ₂ CH ₃	J3
	6 80	NH	CH	2-3	4-Cl, 6-F	CH ₃	J4
	681	NH	CH	2-3 2-3	4-Cl, 6-F	C₂H₅	J4
_	682	NH	- CH	2-3 2-3	4-Cl, 6-F	isopropyl	J4
5	683	NH	CH	2-3 2-3	4-Cl, 6-F	t-butyl	J4
	684	NH	CH	2-3 2-3	4-Cl, 6-F	CF ₃	J4
	68 5	NH	CH CH	2-3 2-3	4-Cl, 6-F	CO ₂ CH ₃	J4
	68 6	NH	CH	2-3 2-3	4-Cl, 6-F	CH ₃	J5
7 0	687	NH	CH	2-3 2-3	4-Cl, 6-F	C ₂ H ₅	J5
10	688	NH NH	CH	2-3	4-Cl, 6-F	isopropyl	J5
	68 9 69 0	NH	CH	2-3	4-Cl, 6-F	t-butyl	J5
	69 1	NH	CH	2-3	4-Cl, 6-F	CF ₃	J5
		NH	CH	2 -3	4-CI, 6-F	CO ₂ CH ₃	J5
1 E	692 69 3	NCH ₃	CH	2-3	4-Cl, 6-F	CF ₃	J1
1 5	69 4	NH	CH	2-3	4-CI	CF ₃	J1
	69 5	CH	NH	1-2	4-Cl, 6-F	CF ₃	J1
	69 6	CH	NCH ₂ C ₆ H ₅		4-Cl, 6-F	CF ₃	J1
	6 97	CH	NCH ₂ CO ₂		4-CI, 6-F	CF ₃	J1
20	69 8	CH	NCOCH ₃	1-2	4-Cl, 6-F	CF ₃	J1
20	69 9	CH	NCH ₂ C≡N		4-Cl, 6-F	CF ₃	J1
	70 0	CH	NH	1-2	4-CI, 6-F	CF ₃	J1
	701	CH	NH	1-2	4-Cl, 6-F	CO ₂ C ₂ H ₅	J1
	702	CH	NH	1-2	4-CI	CO ₂ C ₂ H ₅	J1
2 5	703	N	0	1-2	4-CI, 6-F	CH₃	J7
	704	0	СН	1-2	4-Cl, 6-F	C(CH ₃) ₂ OH	J7
	705	NH	N	2 -3	4-CI, 6-F	CF ₃	J6
	706	NH	N	2-3	4-CI, 6-F	C(CH ₃) ₃	J6
	7 07	NH	N	2 -3	4-Cl, 6-F	CF ₃	J7
30	70 8	NH	N	2-3	4-CI, 6-F	$CH_2C(CH_3)_3$	J1
	70 9	NH	N	2 -3	4-CI, 6-F	3,5-dimethylisoxazolyl	J1
	7 10	NH	N	2- 3	4-CI, 6-F	pyridin-2-yl	J1
	711	NCOCH₃	N	2- 3	4-Cl, 6-F	Н	J1
	712	NH	N	2-3	4-CI, 6-F	C ₇ F ₁₅	J1
35	713	NH	N	2-3	4-Cl, 6-F	CHCl ₂	J1
	714	NH	N	2- 3	4-CI, 6-F	NHCO ₂ C ₂ H ₅	J1
	7 15	NH	N	2 -3	4-CI, 6-F	CH(CH ₃)NHCH ₂ CO ₂ C ₂ H ₅	J1
	716	NH	N	2-3	4-Cl, 6-F	CH(CH ₃)OCOCH ₃	J1
	7 17	NH	N	2-3	4-CI, 6-F	C(CH ₃)=CH ₂	J1
40	718	NH	N	2-3	4-CI, 6-F	CH=C(CH ₃) ₂	J1
	719	NH	N	2 -3	4-CI, 6-F	CH(Br)CH₃	J1
	7 20	NH	N	2-3	6-F	CF ₃	J1
	7 21	NH	N	2 -3	4-CI, 6-F	CH=NC ₆ H ₅	J1
	722	NH	N	2-3	4-CI, 6-F	CH ₂ OCOCH ₃	J1
45	723	NH	N	2 -3	4-CI, 6-F	CH(OCH ₃)C ₆ H ₅	J1
	724	NH	N	2-3	4-Cl, 6-F	CH(OCOCH ₃)C ₆ H ₅	J1 J1
	72 5	NH	N	2-3	4-Cl, 6-F	SCH₃	J 1 J5
	726	NH	N	2 -3	4-Cl, 6-F	C₂H₅ CF₃	J5 J1
	727	NCH ₃	N	2-3	4,6-Cl ₂ 4,6-Cl ₂	CF ₃	J1
50	728	N	NCH₃	2-3	4,0-01 ₂	O F3	0 1

- 44 -

	700	N11.1	NIL.		4-Cl, 6-F	CF ₃	J1
	72 9	NH	NH N	2-3	4,6-Cl ₂	CF ₃	J5
	730	NH	N	2-3 2-3	4-Cl, 6-F	SO ₂ CH ₃	J1
	731	NH NH	. N	2-3 2-3	4-Br,6-F	CF ₃	J1
_	732		, N N	2-3 2-3	4-Br,6-F	C ₂ H ₅	J1
5	733	NH	N N	2 -3	4-CI,6-F	CH₂OH	J1
	734 7 05	NH	N	2-3	4-CI,6-F	C(CH ₃) ₂ OH	J1
	73 5	NH	N	2-3	4-CI,6-F	C(CH ₃)OCH ₂ C ₆ H ₅	J1
	73 6	NH	N	2-3	4-Cl,6-F	SH	J1
7 0	7 37	NH	N	2-3 2-3	4-CI,6-F	SCH(CH₃)C≣N	J1
10	738	NH	N	2-3 2-3	4-Ci,6-F	SC ₂ H ₅	J1
	739	NH NH	N	2-3	4-CI,6-F	SCH ₂ C≡CH	J1
	740	NH	N	2-3	4-CI,6-F	SCH ₂ C ₆ H ₅	J1
	741		N	2-3	4-Cl,6-F	SC≣N	J1
- -	742	NH	N	2 -3	4-CI,6-F	C(CH ₃) ₂ CH ₂ SC≣N	J1
15	743 744	NH NH	N N	2 -3 2 -3	4-CI,6-F	SCH(CH ₃)CO ₂ C ₂ H ₅	J1
	7 44 745	NH	N	2 -3	4-CI,6-F	SCH(CH ₃)CON(CH ₃) ₂	J1
	746	NH	N N	2-3	4-CI,6-F	SCH ₂ C≡CH	J5
	740 74 7	NH	N	2 -3	4-CI,6-F	SCH ₂ CH=CH ₂	J1
20	748	NH	N N	2-3	4-CI,6-F	SCH ₂ C≡N	J1
20	7 4 9	NH	N N	2-3	4-CI,6-F	SCH,C≡CCH,CI	J1
	750	0	CH	2-3	4-CI, 6-F	CH ₂ OCONHCH ₃	J1
	75 1	0	CH	2- 3	4-Cl, 6-F	$CH_2NHCOCH_2(C_8H_4, 2-NO_2)$	J1
	752	0	СН	2-3	4-Cl, 6-F	C(CH ₃)(OH)C ₆ H ₅	J1
2 5	753	0	CH	2-3	4-CI, 6-F	CH ₂ NH ₂	J1
2.0	754	0	CH	2-3	4-CI, 6-F	C(CH ₃)(OH)CH(CH ₃) ₂	J1
	755	Ö	CH	2-3	4-CI, 6-F	CH ₂ NHCOCH ₃	J1
	756	0	CH	2-3	4-CI, 6-F	CH ₂ NHSO ₂ CH ₃	J1
	757	Ö	CH	2-3	4-Cl, 6-F	C(CH ₃) ₂ F	J1
30	758	Ö	CH	2-3	4-CI, 6-F	CH₂CO₂H	J1
20	759	0	СН	2-3	4-CI, 6-F	CH ₂ CON(CH ₃) ₂	J1
	760	0	СН	2 -3	4-CI, 6-F	$CH_2CON(CH_3)(OCH_3)$	J1
	761	0	СН	2-3	4-CI, 6-F	CH₂CONHCH₃	J1
	762	0	СН	2-3	4-ÇI, 6-F	CH ₂ CONH ₂	J1
3 5	763	0	СН	2-3	4-Cl, 6-F	$C_2H_4CON(CH_3)(OCH_3)$	J1
_	764	0	СН	2-3	4-CI, 6-F	C ₂ H ₄ CO ₂ CH ₃	J1
	765	0	СН	2-3	4-CI, 6-F	C₃H _e OH	J1
	766	0	СН	2-3	4-CI, 6-F	C ₂ H ₄ CONHCH ₃	J1
	7 67	NH	N	2-3	4-CI	SCF₃	J1
4 0	76 8	NH	N	2-3	4-CI	CF ₃	J1
	76 9	NH	N	2-3	4-CI	CF ₃	J3

- 45 -

<u>Table 3 Characterizing Data</u>

Melting Points or Physical States of Representative Compounds

	<u>No.</u>	MP/State	No.	MP/State	No.	MP/State	<u>No.</u>	MP/State
5	1	OIL	246	45-9	377	122-30	72 2	117-122 RESIN
_	16	70-72	247	35-8	37 8	200 C >	72 3	107-112 RESIN
	2 5	OIL	248	67-71	379	116-22	724	108-114 RESIN
	26	OIL	249	84-9	380	201-4	72 5	135-140 RESIN
	28	OIL	250	65-68	381	117-24	726	>210
10	30	OIL	251	55-7	382	193-5	727	182-183
	38	246-9	252	OIL	383	131-40	728	174-175
	42	>250	253	GLASS	384	103-5	729	>205
	43	SOLID	254	71-5	385	158-160	730	>205
	49	OIL	255	134-8	386	132-5	731	150-152 RESIN
1 5	96	OIL	256	145-7	387	112-4	732	195 -20 0
	98	>245	257	OIL	388	107-9	733	>205
	99	OIL	258	232-40	39 9	177.5-8.5	734	SOLID
	10 0	OIL	259	165-9	40 5	130	735	118-121 RESIN
	10 1	OIL	260	55-8	46 9	98-100	736	88-92
20	102	OIL	261	65-7	481	SOLID	737	>200
	103	OIL	262	75-7	493	187-8	738	13 3-135
	104	OIL	263	>50	500	208-10	739	130-132
	105	>250	264	155-7	513	178-181	740	178-180
	10 6	OIL	265	130- 6	522	78-8 0	741	118-121 RESIN
2 5	107	OIL	266	258-61	527	152-154	742	150-155
	108	>250	267	110-8	56 3	165 -16 6	743	SOLID
	10 9	OIL	268	73 -7	59 5	>240	744	160-162
	110	OIL	269	270 -5	61 8	235-237.5	745	>200
	112	86-88	270	265-72	693	60-65	746	106-109
30	2 21	193.5-6	271	62-7 2	694	221. 5-22 3	747	98-100
	22 2	183-6	272	OIL	69 5	160-162	74 8	104-110 RESIN
	22 3	OIL	273	220-2.5	69 6	173-177	74 9	155-158 RESIN
	2 24	OIL	274	116 SOFTENS	697	60-6 3	750	137-139
	22 5	OIL	275	OIL	69 8	142-145.5	751	189-190
35	22 6	63-6	276	145-53	699	95-102	752	78-82
	2 27	134-6	277	17 9- 82	700	160-162	753	87-89
	22 8	42-5	278	18 9- 92	701	245-248	754	75-77
	22 9	OIL	279	197-8	70 2	258-260	755	96-98
	2 30	163 -5	280	215-6	70 5	102-103	756	90-92 60-62
40	231	65-70	362	152-8	70 6	88-89	757	
	2 32	186-91	363	>165	708	140 DEC	758 750	95-97 144-146
	23 3	85-9 0	364	SOLID	709	>200	759 760	146-147
	2 34	65-7 0	365	172-7	710	130 RESIN	760 764	70-76
	23 5	63-7	366	130	711	>200	761	185-187
45	23 6	56- 8	367	150- 5	712	93-98 RESIN	762	63-65
	237	141-2	368	87-93	713	123-130 RESIN	763 764	OIL
	23 8	143 -5	369	125-30	714	160-165 RESIN	764 765	50-54
	2 39		370	130	715	90-95	765 766	172-173
<u></u> -	24 0		371	SOLID	716	115-120 RESIN	76 6 76 7	239-241
50	241	67-7 0	372	SOLID	717	120-125	101	200-2-1
	24 2		373	160	718	110-116		
	24 3	51-5 5	374	190	719	120-125		

PCT/US98/03647

.- 46 -

No.	MP/State	<u>No.</u>	MP/State	No.	MP/State	No.	MP/State
244	OIL	375	>200	72 0	128-132 RESIN	-	-
24 5	OIL	376	142-8	721	145-150		

5 Biological Testing

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The benzofused heterocyclic compounds of this invention were tested for pre- and postemergence herbicidal activity using a variety of crops and weeds. The test plants included soybean (Glycine max var. Winchester), field corn (Zea mays var. Pioneer 3732), wheat (Triticum aestivum var. Lew), morningglory (Ipomea lacunosa or Ipomea hederacea), velvetleaf (Abutilon theophrasti), green foxtail (Setaria viridis), Johnsongrass (Sorghum halepense), blackgrass (Aloepecurus myosuroides), common chickweed (Stellaria media), and common cocklebur (Xanthium strumarium L.).

For preemergence testing, two disposable fiber flats (8 cm x 15 cm x 25 cm) for each rate of application of each candidate herbicide were filled to an approximate depth of 6.5 cm with steam-sterilized sandy loam soil. The soil was leveled and impressed with a template to provide five evenly spaced furrows 13 cm long and 0.5 cm deep in each flat. Seeds of soybean, wheat, corn, green foxtail, and johnsongrass were planted in the furrows of the first flat, and seeds of velvetleaf, morningglory, common chickweed, cocklebur, and blackgrass were planted in the furrows of the second flat. The five-row template was employed to firmly press the seeds into place. A topping soil of equal portions of sand and sandy loam soil was placed uniformly on top of each flat to a depth of approximately 0.5 cm. Flats for postemergence testing were prepared in the same manner except that they were planted 9-14 days prior to the preemergence flats and were placed in a greenhouse and watered, thus allowing the seeds to germinate and the foliage to develop.

In both pre- and postemergence tests, a stock solution of the candidate herbicide was prepared by dissolving 0.27g of the compound in 20 mL of water/acetone (50/50) containing 0.5% v/v sorbitan monolaurate. For an application rate of 3000 g/ha of herbicide a 10 mL portion of the stock solution was diluted with water/acetone (50/50) to 45 mL. The volumes of stock solution and

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diluent used to prepare solutions for lower application rates are shown in the following table:

5	Application Rate (g/ha)	Volume of Stock Solution (mL)	Volume of Acetone/Water (mL)	Total Volume of Spray Solution (mL)
	3000	10	35	45
	1000	3	42	45
	300	1	44	45
	100	0.3	45	45.3
10	30	0.1	45	45.1
10	10	0.03	45	45.03
	3	0.01	45	45.01

The preemergence flats were initially subjected to a light water spray.

The four flats were placed two by two along a conveyor belt (i.e., the two preemergence followed by the two postemergence flats). The conveyor belt fed under a spray nozzle mounted about ten inches above the postemergent foliage. The preemergent flats were elevated on the belt so that the soil surface was at the same level below the spray nozzle as the foliage canopy of the postemergent plants. The spray of herbicidal solution was commenced and once stabilized, the flats were passed under the spray at a speed to receive a coverage equivalent of 1000L/ha. At this coverage the application rates are those shown in the above table for the individual herbicidal solutions. The preemergence flats were watered immediately thereafter, placed in the greenhouse and watered regularly at the soil surface. The postemergence flats were immediately placed in the green-house and not watered until 24 hours after treatment with the test solution. Thereafter they were regularly watered at ground level. After 12-17 days the plants were examined and the phytotoxicity data were recorded.

Herbicidal activity data at selected application rates are given for various compounds of this invention in Table 4 and Table 5. The test compounds are identified by numbers which correspond to those in Tables 1 and 2.

Phytotoxicity data were taken as percent control. Percent control was determined by a method similar to the 0 to 100 rating system disclosed in

"Research Methods in Weed Science," 2nd ed., B. Truelove, Ed.; Southern Weed Science Society; Auburn University, Auburn, Alabama, 1977. The rating system is as follows:

Herbicide Rating System

5	Rating. Percent Control	Description of Main Categories	Crop Description	Weed Description
	0	No effect	No crop reduction/injury	No weed control
10	10		Slight dis- coloration or stunting	Very poor weed control
	20	Slight effect	Some discoloration, stunting or stand loss	Poor weed control
1 5	30		Crop injury more pronounced but not lasting	Poor to defi- cient weed control
20	40		Moderate injury, crop usually recovers	Deficient weed control
	50	Moderate effect	Crop injury more lasting, recovery doubtful	Deficient to moderate weed control
25	60		Lasting crop injury, no recovery	Moderate weed control
	70		Heavy injury and stand loss satisfactory	Control some- what less than
30	80	Severe	Crop nearly destroyed a few survivors	Satisfactory to weed control
	90		Only occasional live plants left	Very good to excellent control
3 5	100	Complete effect	Complete crop destruction	Complete weed destruction

Formulation

The compounds of the present invention were tested in the laboratory as water/acetone (50/50) solutions containing 0.5% v/v sorbitan monolaurate

emulsifier. It is expected that all formulations normally employed in applications of herbicides would be usable with the compounds of the present invention. These include wettable powders, emulsifiable concentrates, water suspensions, flowable concentrates, and the like.

Table 4. PREEMERGENCE HERBICIDAL ACTIVITY (% CONTROL)

	<u>No.</u>	SOY	<u>WHT</u>	CRN	<u>ABUTH</u>	<u>IPOSS</u>	STEME	XANPE	ALOMY	<u>SETVI</u>	SORHA
	1	10 0	85	90	10 0	100	100	100	90	100	95
	16	1 0 0	70	90	10 0	100	100	90	80	100	9 5
	2 5	10 0	100	100	10 0	100	100	95	90	100	100
10	2 6	10 0	90	90	10 0	100	10 0	100	9 5	10 0	100
	2 8	100	10 0	9 5	10 0	100	10 0	100	10 0	100	100
	30	10 0	10 0	95	10 0	100	100	90	10 0	100	100
	38	6 0	50	80	10 0	100	0	70	30	75	60
	42	0	10	0	100	60	30	20	50	30	0
15	43	50	40	80	10 0	100	10		6 0	70	80
	49	9 5	50	80	10 0	100	20	90		100	90
	9 6	10 0	90	9 5	10 0	100	100		9 0	100	95
	9 8	5 0	40	80	80	75	70	60	10	30	65
	9 9	40	50	6 0	10 0	10 0	100		6 0	100	65
20	10 0	40	3 0	80	10 0	100	20		6 0	50	70
	101	80	7 0	100	10 0	100		80	80	100	100
	102	20	30	10	10 0	70		50	90	100	60
	103	50	50	80	10 0	100		70	90	10 0	70
	104	10 0	10 0	100	10 0	100		100	10 0	100	100
25	10 6	30	4 0	70	10 0	10 0	9 5	60	70	90	5 5
	107	80	6 0	90	10 0	10 0	100	40	7 5	100	100
	108	0	0	10	7 0	50	40	10	50	50	30
	10 9	10 0	10 0	90	10 0	10 0	100	100	10 0	100	100
	110	10 0	50	70	10 0	90	100	40	80	100	100
30	112	10 0	10 0	10 0	10 0	10 0	100	100	100	100	100
	221	70	6 0	85	100	100	80	ND	ND	100	95 400
	22 2	100	70	90	10 0	100	100	100	ND	100	100 100
	22 3	100	50	80	10 0	100	100	90	ND 00	100	100 100
	224	10 0	80	90	10 0	100	100	95	80 ND	100 100	60
35	22 5	4 0	20	30	90	50	70	50	ND		80
	22 6	70	50	7 0	100	90	90	60	ND	100 100	100
	22 7	10 0	80	90	10 0	100	100	ND	9 5	100	100
	22 8	100	80	95	10 0	100	100	90	ND 80	100 100	100
	22 9	10 0	7 0	90	10 0	10 0	100	95	80		100
4 0	23 0	100	40	80	10 0	100	100	100	80	100 100	100
	231	10 0	80	100	10 0	100	10 0	100	90 ND	40	25
	23 2	20	30	50	90	80	20	10	ND ND	60	25 50
	23 3	40	30	70	100	95	20	20	ND 80	100	100
	234	10 0	100	100	10 0	10 0	10 0	10 0	80	100	100

- 50 -

	235	100	90	100	10 0	100	100	100	80	100	100
	23 6	100	7 0	95	10 0	100	100	100	. 80	100	100
	237	100	90	90	100	100	100	100	10 0	100	100
	238	100	60	70	10 0	100	6 0	80	50	90	90
5	239	1 0 0	70	90	10 0	100	10 0	ND	ND	100	90
	240	10 0	9 5	95	10 0	100	10 0	100	ND	100	100
	241	60	70	95	10 0	10 0	1 0 0	100	ND	100	100
	242	10 0	100	10 0	10 0	10 0	10 0	10 0	10 0	100	100
	243	10 0	80	95	10 0	10 0	10 0	100	ND	100	100
10	244	95	80	100	10 0	90	70	100	7 0	100	100
	24 5	10 0	6 0	80	10 0	100	9 0	100	70	100	80
	246	10 0	10 0	100	10 0	100	100	10 0	10 0	100	100
	247	100	90	90	10 0	100	9 5	100	8 5	100	100
	24 8	10 0	90	95	10 0	10 0	10 0	100	9 5	100	100
15	24 9	10 0	80	95	10 0	100	100	90	80	10	100
	25 0	80	40	50	10 0	10 0	ND	100	6 0	100,	70
	251	90	90	95	10 0	100	95	100	90	100	100
	25 2	10 0	10 0	10 0	10 0	10 0	100	ND	10 0	100	100
	25 3	10 0	9 5	100	10 0	10 0	100	ND 50	ND	100	100
20	254	25	20	80	100	50	30	50	6 0	100	80 10 0
	25 5	10 0	90	95	10 0	100	100	100	N D 7 0	100 100	90
	25 6	10 0	80	95	10 0	1 0 0	100	ND 20	2 0	70	10
	257	40	0	10	90	70	0	60	ND	40	40
	25 8	30	30	7 5	100	60	0 10 0	55	ND ND	100	95
25	259	70	40	80	100	70	100	100	95	100	100
	26 0	100	70	80	100	100 1 0 0	100 100	90	8 0	100	100
	261	100	80	95 40	100 100	100	1 0 0	100	5 0	100	70
	26 2	90	40	4 0	10 0 10 0	100 100	100	95	7 5	100	70
2.0	26 3	100	50	65 10	20	0	20	30	0	10	10
30	264	0	0	80	9 0	1 0 0	20	7 0	ND	80	60
	26 5	7 0	4 0 3 0	60	40	70	0	0	ND	30	30
	26 6 26 7	5 0 0	10	20	10	10	0	50	50	5	0
	267 268	30	30	50	1 0 0	95	20	0	ND	60	6 0
35	26 9	60	30	80	10 0	10 0	100	10 0	7 0	100	75
33	27 0	70	70	90	100	10 0	ND	60	6 5	100	100
	271	80	70	90	10 0	100	10 0	100	8 0	10 0	90
	27 2	20	0	20	10 0	7 0	10 0	20	70	90	60
	27 3	100	80	90	10 0	10 0	10 0	100	90	100	100
4 0	274	10 0	10 0	90	10 0	100	90	100	9 5	100	100
	27 5	10 0	80	10 0	10 0	100	100	100	8 0	100	95
	3 62	10 0	10 0	100	10 0	100	10 0	10 0	10 0	100	100
	36 3	10 0	10 0	10 0	10 0	100	10 0	10 0	10 0	100	100
	3 64	10 0	6 0	80	10 0	10 0	10 0	100	8 0	100	80
45	36 5	ND	30	30	10 0	10 0	10 0	100	60	75	60
	36 6	10	10	0	70	20	0	10	0	50	40
	36 7	10 0	95	10 0	10 0	10 0	100	100	90	100	100
	36 8	10 0	10 0	95	10 0	10 0	10 0	100	9 5	100	100
	36 9	10 0	10 0	10 0	100	10 0	10 0	100	10 0	100	100
50	37 0	10 0	10 0	9 5	10 0	100	10 0	100	90	100	100
	371	10 0	95	100	10 0	100	100	100	10 0	100	100
	37 2	10 0	9 0	95	10 0	10 0	10 0	10 0	8 0	100	80

- 51 -

	37 3	100	7 0	90	10 0	100	100	100	7 0	100	90
	374	30	0	10	100	95	90	80	4 0	100	75
	375	80	30	90	100	80	95	80	80	100	95
	376	50	- 6 0	80	10 0	100	100	100	10 0	100	70
5	377	100	7 0	90	10 0	100	ND	10 0	10 0	100	100
•	378	90	70	90	10 0	10 0	10 0	100	8 0	100	95
	379	100	5 0	7 0	10 0	10 0	ND	100	80	100	95
	380	80	3 5	20	10 0	100	ND	80	9 0	100	70
	381	100	40	80	10 0	100	ND	10 0	90	100	80
10	382	60	45	30	100	70	ND	60	90	95	80
	383	80	40	20	10 0	6 0	ND	70	80	7 5	5 5
	39 9	95	80	95	100	95	100	70	6 0	100	0
	493	80	70	90	10 0	100	100	70	75	10 0	100
	500	95	75	90	10 0	10 0	100	100	7 5	100	100
15	522	90	40	80	10 0	10 0	100	50	75	100	100
13	59 5	10	0	0	60	50	10	20	ND	0	40

Rate of Application is 0.3 Kg/Ha. SOY is soybean; WHT is wheat; CRN is corn; ABUTH is velvetleaf; IPOSS is morningglory; STEME is chickweed; XANPE is cocklebur; ALOMY is blackgrass, SETVI is green foxtail; SORHA is johnsongrass

20 Table 5. POSTEMERGENCE HERBICIDAL ACTIVITY (% CONTROL)

	No.	SOY	WHT	CRN	<u>ABUTH</u>	<u>IPOSS</u>	STEME	XANPE	<u>ALOMY</u>	SETVI	SORHA
										•	00
	.1	9 5	6 5	80	10 0	100	9 0	100	70	80	80
	16	9 5	6 0	80	10 0	100	7 0	95	70	80	80
	25	10 0	80	90	10 0	100	10 0	100	8 0	100	90
25	2 6	9 6	6 0	80	10 0	100	8 0	100	8 0	100	80
	2 8	10 0	80	80	10 0	100	10 0	90	10 0	100	9 5
	3 0	9 5	80	90	10 0	100	10 0	1 0 0	90	100	100
	38	70	35	60	10 0	100	0	45	20	40	50
	42	6 5	30	60	90	60		50	40	100	20
30	43	80	30	70	10 0	100	70	50		50	50
	49	9 5	7 0	80	10 0	100	40	30		100	90
	96	100	9 0	90	100	100	10 0	100		100	100
	98	40	10	50	6 0	20	5	20	5	40	20
	99	80	40	80	10 0	100	95	70		70	65
35	100	85	40	60	90	100	50	50		30	40
	101	95	50	80	10 0	100			60	65	65
	102	80	30	75	10 0	100			60	90	60
	10 3	90	50	80	10 0	80		80	7 0	100	60
	104	100	100	100	10 0	100				100	100
40	10 ⁻⁷	80	30	75	100	100			60	100	70
-20	107	95	40	100	10 0	100	100		9 0	10 0	100
	107 108	5 0	20	60	20	60	0	10	10	70	20

- 52 -

	100	90	90	80	10 0	100		100	90	100	90
	10 9 11 0	8 0	4 0	50	1 0 0	100		100	. 70	80	70
	112	10 0	10 0	100	1 0 0	100	10 0	10 0	100	100	100
	221	95	50-		10 0	100	100	60	40	70	70
5	222	100	70	90	1 0 0	100	100	1 0 0	100	10 0	100
J	223	95	40	90	10 0	100	100	100	ND	10 0	100
	224	9 5	70	100	1 0 0	100	100	100	90	10 0	ND
	22 5	60	30	60	100	75	ND	70	ND	90	60
	22 6	70	40	80	100	95	80	90	ND	100	80
10	227	95	6 0	90	100	100	100	10 0	100	10 0	100
10	228	90	50	80	10 0	100	80	95	ND	100	90
	22 9	9 5	60	80	10 0	100	100	100	70	10 0	100
	23 0	95	40	80	100	100	90	100	70	100	90
	231	100	70	100	10 0	100	100	ND	100	100	100
15	23 2	75	50	30	100	80	20	40	ND	30	10
	23 3	90	30	60	10 0	100	30	30	ND	30	30
	234	100	10 0	100	10 0	100	100	10 0	100	10 0	100
	23 5	10 0	10 0	10 0	10 0	100	10 0	10 0	100	10 0	100
	236	100	7 5	90	10 0	100	100	100	80	10 0	100
20	23 7	10 0	9 5	100	10 0	100	ND	100	100	10 0	100
	23 8	80	30	7 0	10 0	100	ND	100	40	80	70
	23 9	9 5	6 0	80	10 0	100	100	100	ND	100	80
	24 0	9 5	9 5	100	10 0	100	10 0	100	ND	100	100
	241	9 0	6 0	70	10 0	100	85	9 5	ND	100	70
25	24 2	10 0	10 0	100	10 0	100	10 0	10 0	100	100	100
	24 3	9 5	70	95	10 0	100	10 0	10 0	ND	100	100
	244	9 5	60	90	10 0	100	100	10 0	7 5	10 0	ND
	24 5	8 5	4 0	7 5	10 0	100	60	70	50	70	70
	24 6	9 5	10 0	100	10 0	100	100	100	ND	100	100
30	24 7	9 5	80	100	10 0	100	100	10 0	100	100	ND
	24 8	8 0	50	9 5	10 0	100	10 0	100	ND	10 0	100
	24 9	9 5	80	100	10 0	100	100	100	100	1 0 0	ND
	25 0	9 5	5 0	80	10 0	100	80	100	40	100	100 95
	25 1	9 5	70	90	10 0	100	100	95	100	100	100
35	25 2	9 5	9 0	10 0	100	100	100	ND 100	100	100 100	100
	25 3	9 5	10 0	10 0	100	100	100	100	ND 5 0	100	80
	25 4	9 5	40	80	100	70	ND 100	95 400		100 100	100
	25 5	10 0	100	100	10 0	100	100	100 100	N D 8 0	1 0 0	100
	25 6	10 0	80	90	100	100	100 30	70	30	70	50
40	257	70	20	70	10 0	60 70	30 5	50	30	60	50
	25 8	80	3 0	60	80	70 90	30	70	55	80	70
	25 9	80	35	7 5	10 0 10 0	100	ND	100	90	100	90
	26 0	90	80	70	100 100	100	100	100	10 0	100	100
4 =	26 1	95	80	10 0	100 100	100	95	95	50	100	80
45	26 2	9 5	60	80	100 100	100	100	100	60	90	70
	26 3	9 5	80	90	40	40	0	30	0	ND	20
	2 64	5 0	20	50 60	100	100	30	20	ND	40	20
	26 5	7 0	40	60 60	50	60	10	10	ND	40	40
ΕO	26 6	6 0	4 0	50	80	40	10	10	20	30	20
50	26 7	50 70	15 40	60	5 0	90	20	ND	ND	70	40
	26 8 26 9	7 0 9 0	4 0 4 0	70	1 0 0	80	80	ND	ND	70	60
	209	90	40	,,	100	50					

PCT/US98/03647

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	270	7 0	40	50	100	60	40	ND	5 0	50	50
	271	80	40	60	10 0	100	100	ND	, ND	70 -	50
	27 2	50	30	4 5	100	60	50	50	20	70	40
	27 3	9 5	60-	95	100	100	90	100	80	10 0	100
5	274	9 5	6 0 -	95	10 0	100	90	10 0	90	100	100
	27 5	10 0	7 0	90	10 0	100	100	100	9 5	10 0	100
	36 2	10 0	10 0	10 0	10 0	100	100	ND	1 0 0	100	100
	36 3	10 0	10 0	100	10 0	10 0	ND	10 0	10 0	10 0	100
	364	9 5	4 0	80	10 0	100	100	1 0 0	ND	10 0	100
10	36 5	10 0	40	7 0	10 0	100	10 0	ND	70	80	30
	36 6	70	30	80	95	80	30	100	30	50	50
	36 7	10 0	10 0	100	10 0	100	100	100	100	100	100
	36 8	10 0	10 0	100	10 0	100	100	10 0	100	100	100
	36 9	10 0	80	100	10 0	100	ND	100	100	100	100
1 5	37 0	10 0	95	100	10 0	100	100	100	100	100	100
	371	95	10 0	100	10 0	100	10 0	10 0	ND	100	100
	3 72	10 0	10 0	10 0	10 0	10 0	10 0	100	100	100	100
	37 3	10 0	80	10 0	10 0	100	10 0	100	10 0	10 0	100
	374	80	2 5	30	100	9 5	80	100	25	80	60
20	37 5	9 5	4 0	90	10 0	95	100	100	90	80	100
	37 6	9 0	5 0	95	10 0	10 0	ND	100	90	10 0	100
	37 7	9 5	8 0	100	10 0	100	ND	100	10 0	10 0	100
	37 8	90	40	90	10 0	90	ND	100	80	10 0	100
	37 9	9 5	80	10 0	10 0	100	ND	10 0	70	10 0	100
25	38 0	9 5	30	95	10 0	10 0	ND	100	70	100	80
	381	9 5	4 0	95	10 0	10 0	ND	10 0	10 0	100	100
	38 2	80	4 0	100	10 0	10 0	ND	100	80	90	80
	38 3	9 5	4 0	9 5	100	95	ND	100	60	95	70
	39 9	9 5	30	70	10 0	10 0	10 0	10 0	50	70	60
30	49 3	9 5	6 0	90	10 0	100	80	100	6 5	100	100
	50 0	9 5	65	9 5	10 0	10 0	90	100	80	100	100
	52 2	90	4 5	90	10 0	100	100	10 0	50	100	100
	59 5	50	10	60	30	40	0	20	10	20	20

Rate of Application is 0.3 Kg/Ha. SOY is soybean; WHT is wheat; CRN is corn; ABUTH is velvetleaf; IPOSS is morningglory; STEME is chickweed; XANPE is cocklebur; ALOMY is blackgrass, SETVI is green foxtail; SORHA is johnsongrass

Herbicidal compositions are prepared by combining herbicidally effective amounts of the active compounds with adjuvants and carriers normally employed in the art for facilitating the dispersion of active ingredients for the particular utility desired, recognizing the fact that the formulation and mode of application of a toxicant may affect the activity of the material in a given application. Thus, for agricultural use the present herbicidal compounds may be formulated as granules of relatively large particle size, as water-soluble or water-dispersible granules, as powdery dusts, as wettable powders, as emulsifiable concentrates, as solutions, or

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as any of several other known types of formulations, depending on the desired mode of application. It is to be understood that the amounts specified in this specification are intended to be approximate only, as if the word "about" were placed in front of the amounts specified.

These herbicidal compositions may be applied either as water-diluted sprays, or dusts, or granules to the areas in which suppression of vegetation is desired. These formulations may contain as little as 0.1%, 0.2% or 0.5% to as much as 95% or more by weight of active ingredient.

Dusts are free flowing admixtures of the active ingredient with finely divided solids such as talc, natural clays, kieselguhr, flours such as walnut shell and cottonseed flours, and other organic and inorganic solids which act as dispersants and carriers for the toxicant; these finely divided solids have an average particle size of less than about 50 microns. A typical dust formulation useful herein is one containing 1.0 part or less of the herbicidal compound and 99.0 parts of talc.

Wettable powders, also useful formulations for both pre- and postemergence herbicides, are in the form of finely divided particles which disperse
readily in water or other dispersant. The wettable powder is ultimately applied to the
soil either as a dry dust or as an emulsion in water or other liquid. Typical carriers for
wettable powders include Fuller's earth, kaolin clays, silicas, and other highly
absorbent, readily wet inorganic diluents. Wettable powders normally are prepared
to contain about 5-80% of active ingredient, depending on the absorbency of the
carrier, and usually also contain a small amount of a wetting, dispersing or
emulsifying agent to facilitate dispersion. For example, a useful wettable powder
formulation contains 80.0 parts of the herbicidal compound, 17.9 parts of Palmetto
clay, and 1.0 part of sodium lignosulfonate and 0.3 part of sulfonated aliphatic
polyester as wetting agents. Additional wetting agent and/or oil will frequently be
added to the tank mix for post-emergence application to facilitate dispersion on the
foliage and absorption by the plant.

Other useful formulations for herbicidal applications are emulsifiable concentrates (ECs) which are homogeneous liquid compositions dispersible in water or other dispersant, and may consist entirely of the herbicidal compound and a liquid or solid emulsifying agent, or may also contain a liquid carrier, such as xylene, heavy

-WO 98/38188 PCT/US98/03647

aromatic naphthas, isphorone, or other non-volatile organic solvents. For herbicidal application these concentrates are dispersed in water or other liquid carrier and normally applied as a spray to the area to be treated. The percentage by weight of the essential active ingredient may vary according to the manner in which the composition is to be applied, but in general comprises 0.5 to 95% of active ingredient by weight of the herbicidal composition.

Flowable formulations are similar to ECs except that the active ingredient is suspended in a liquid carrier, generally water. Flowables, like ECs, may include a small amount of a surfactant, and will typically contain active ingredients in the range of 0.5 to 95%, frequently from 10 to 50%, by weight of the composition. For application, flowables may be diluted in water or other liquid vehicle, and are normally applied as a spray to the area to be treated.

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Typical wetting, dispersing or emulsifying agents used in agricultural formulations include, but are not limited to, the alkyl and alkylaryl sulfonates and sulfates and their sodium salts; alkylaryl polyether alcohols; sulfated higher alcohols; polyethylene oxides; sulfonated animal and vegetable oils; sulfonated petroleum oils; fatty acid esters of polyhydric alcohols and the ethylene oxide addition products of such esters; and the addition product of long-chain mercaptans and ethylene oxide. Many other types of useful surface-active agents are available in commerce. Surface-active agents, when used, normally comprise 1 to 15% by weight of the composition.

Other useful formulations include suspensions of the active ingredient in a relatively non-volatile solvent such as water, corn oil, kerosene, propylene glycol, or other suitable solvents.

Still other useful formulations for herbicidal applications include simple solutions of the active ingredient in a solvent in which it is completely soluble at the desired concentration, such as acetone, alkylated naphthalenes, xylene, or other organic solvents. Granular formulations, wherein the toxicant is carried on relative coarse particles, are of particular utility for aerial distribution or for penetration of cover crop canopy. Pressurized sprays, typically aerosols wherein the active ingredient is dispersed in finely divided form as a result of vaporization of a low boiling dispersant solvent carrier, such as the Freon fluorinated hydrocarbons, may

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also be used. Water-soluble or water-dispersible granules are free-flowing, non-dusty, and readily water-soluble or water-miscible. The soluble or dispersible granular formulations described in US 3,920,442 are useful herein with the present herbicidal compounds. In use by the farmer on the field, the granular formulations, emulsifiable concentrates, flowable concentrates, solutions, etc., may be diluted with water to give a concentration of active ingredient in the range of say 0.1% or 0.2% to 1.5% or 2%.

The active herbicidal compounds of this invention may be formulated and/or applied with insecticides, fungicides, nematicides, plant growth regulators, fertilizers, or other agricultural chemicals and may be used as effective soil sterilants as well as selective herbicides in agriculture. In applying an active compound of this invention, whether formulated alone or with other agricultural chemicals, an effective amount and concentration of the active compound is of course employed; the amount may be as low as, e.g. about 1 to 250 g/ha, preferably about 4 to 30 g/ha. For field use, where there are losses of herbicide, higher application rates (e.g., four times the rates mentioned above) may be employed.

The active herbicidal compounds of the present invention may also be used in combination with other herbicides. Such herbicides include, for example: N-(phosphonomethyl) glycine ("glyphosate"); aryloxyalkanoic acids such as (2,4dichlorophenoxy)acetic acid ("2,4-D"), (4-chloro-2-methylphenoxy)acetic acid ("MCPA"), (+/-)-2-(4-chloro-2-methylphenoxy)propanoic acid (MCPP); ureas such as N,N-dimethyl-N'-[4-(1-methylethyl)phenyl]urea ("isoproturon"); imidazolinones such 2-[4,5-dihydro-4-methyl-4-(1-methylethyl)-5-oxo-1H-imidazol-2-yl]-3pyridinecarboxylic acid ("imazapyr"), a reaction product comprising (+/-)-2-[4,5dihydro-4-methyl-4-(1-methylethyl)-5-oxo-1H-imidazol-2-yl]-4-methylbenzoic acid and (+/-)-2-[4,5-dihydro-4-methyl-4-(1-methylethyl)-5-oxo-1H-imidazol-2-yl]-5-(+/-)-2-[4,5-dihydro-4-methyl-4-(1-("imazamethabenz"), methylbenzoic acid methylethyl)-5-oxo-1H-imidazol-2-yl]-5-ethyl-3-pyridinecarboxylic acid (+/-)-2-[4,5-dihydro-4-methyl-4-(1-methylethyl)-5-oxo-1H-("imazethapyr"). and imidazol-2-yl]-3-quinolinecarboxylic acid ("imazaquin"); diphenyl ethers such as 5-[2chloro-4-(trifluoromethyl)phenoxy]-2-nitrobenzoic acid ("acifluorfen"), methyl 5-(2,4-5-[2-chloro-4-(trifluoro-("bifenox"), and dichlorophenoxy)-2-nitrobenzoate

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hydroxymethyl)phenoxy]-N-(methylsulfonyl)-2-nitrobenzamide ("fomasafen"); benzonitriles such as 4-hydroxy-3,5-diiodobenzonitrile ("ioxynil") and 3,5-dibromo-4hydroxybenzonitrile ("bromoxynil"); sulfonylureas such as 2-[[[[(4-chloro-6-methoxy-2pyrimidinyl)amino]carbonyl]amino]sulfonyl]benzoic acid ("chlorimuron"), 2-chloro-N-[[(4-methoxy-6-methyl-1,3,5-triazin-2-yl)amino]carbonyl]benzenesulfonamide 2 - [[[[(4 , 6 - d i m e t h o x y - 2 -("chlorsulfuron"), pyrimidinyl)amino]carbonyl]amino]sulfonyl]methyl]benzoic acid ("bensulfuron"), 2-[[[[(4,6-dimethoxy-2-pyrimidinyl)amino]carbonyl]amino]sulfonyl]-1-methyl-1H-pyrazolacid ("pyrazosulfuron"), 3-[[[[(4-methoxy-6-methyl-1,3,5-triazin-2-4-carboxylic yl)amino]carbonyl]amino]sulfonyl]-2-thiophenecarboxylic acid ("thifensulfuron"), and 2-(2-chloroethoxy)-N-[[(4-methoxy-6-methyl-1,3,5-triazin-2yl)amino]carbonyl]benzenesulfonamide ("triasulfuron"); 2-(4-aryloxyphenoxy)alkanoic acids such as (+/-)-2-[4-[(6-chloro-2-benzoxazolyl)oxy]phenoxy]propanoic acid ("fenoxaprop"), (+/-)-2-[4-[[5-(trifluoromethyl)-2-pyridinyl]oxy]phenoxy]propanoic acid (+/-)-2-[4-(6-chloro-2-quinoxalinyl)oxy]phenoxy]propanoic ("fluazifop"), 15 ("quizalofop"), and (+/-)-2-[-(2,4-dichlorophenoxy)phenoxy]propanoic acid ("diclofop"); benzothiadiazinones such as 3-(1-methylethyl)-1H-2,1,3-benzothiadiazin-4(3H)-one 2,2-dioxide ("bentazone"); 2-chloroacetanilides such as N-butoxymethyl)-2-chloro-2-chloro-N-(2-ethyl-6-methylphenyl)-N-(2-("butachlor"); 2',6'-diethylacetanilide methoxy-1-methylethyl)acetamide ("metachlor"), 2-chloro-N-(ethoxymethyl)-N-(2-20 ethyl-6-methylphenyl)acetamide ("acetochlor"), and (RS)-2-chloro-N-(ethoxymethyl)-N-(2-methoxy-1-methylethyl)acetamide ("dimethenamide"); arenecarboxylic acids such as 3,6-dichloro-2-methoxybenzoic acid ("dicamba"); and pyridyloxyacetic acids such as [(4-amino-3,5-dichloro-6-fluoro-2-pyridinyl)oxy]acetic acid ("fluroxypyr").

It is apparent that various modifications may be made in the formulations and application of the compounds of the present invention without departing from the inventive concepts herein, as defined in the claims.

We claim:

1. A compound having the formula

where

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- (1) A is nitrogen double-bonded to position 2 and B is oxygen;
- (2) A is oxygen and B is CR¹ double bonded to position 2;
- (3) A is NH and B is nitrogen double-bonded to position 2;
- (4) A is nitrogen double bonded to position 2 and B is NR²;
- (5) A is CH double bonded to position 2 and B is NR²;
- (6) A is NH and B is CR¹ double bonded to position 2; or
- (7) A and B are NH;

R is hydrogen, hydroxy, mercapto, straight or branched chain lower alkyl, cycloalkyl, alkoxy, aryl, heteroaryl, alkenyl, haloalkyl, hydroxyalkyl, haloaryl, alkoxyaryl, arylalkyl, aryloxyalkyl, haloarylalkyl, alkylthio, heterocyclyl, alkoxyalkyl, arylcarbonyloxyalkyl, alkylcarbonyloxyalkyl, alkoxylalkyloxyalkyl, aminocarbonyloxyalkyl, aminoalkyl, cyanoalkyl, aminoalkenyl, carboxy, carboxyalkyl, alkylcarboxy, alkylcarboxyalkyl, formyl, aminocarbonyl, amino, oxygen, cyano, nitro, alkylsulfonylamino, alkylcarboxyoxyalkyl, aminosulfonyl, alkylsulfonyl, alkoxycarbonylalkylaminoalkyl, alkoxycarbonylamino, akylcarboxylalkoxy, (aryl)(alkylcarbonyloxy)alkyl, arylalkoxyalkyl, (aryl)(alkoxy)alkyl, aryliminoalkyl, cyanothioalkyl, cyanothio, alkynylalkylthio, arylalkylthio, cyanoalkylthio, alkenylalkylthio, aminocarbonylalkylthio, alkoxycarbonylalkylthio, arylalkylcarbonylaminoalkyl, aminocarbonyloxyalkyl, haloalkylalkynylalkylthio, alkylsulfonylaminoalkyl, alkylcarbonylaminoalkyl, (hydroxy)(aryi)alkyl, aminocarbonylalkyl, alkoxycarbonyl, and alkenyloxy, where the amino group may be substituted with one or two substituents independently selected from alkyl, hydroxy, alkoxy, carboxy, aryl, alkylsufonyl or haloalkylsulfonyl;

R¹ is hydrogen, lower alkyl, or haloalkyl;

 R^2 is hydrogen, alkyl, haloalkyl, $CO_2(alkyl)$, $CH_2CO_2(alkyl)$, $CH_2CONH(alkyl)$, $CH_2CON(alkyl)_2$, CH_2CO_2H , CH_2OCH_3 , $SO_2(alkyl)$, $CH_2CH=CH_2$, or $CH_2C=CH$;

X is selected from hydrogen, F, Cl, Br, alkyl, haloalkyl, CN, NO₂, and

5 **NH₂**;

n is 0-3;

J is selected from

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 $\label{eq:ch2} \text{and} \quad \mathsf{R}^3 \quad \text{is selected from hydrogen, alkyl, haloalkyl, } \mathsf{CH_2CN}, \\ \mathsf{CH_2CH=CH_2}, \ \mathsf{CH_2C=CH}, \ \mathsf{CH_2CO_2(alkyl)}, \ \mathsf{CH_2OCH_3}, \ \text{and} \ \mathsf{NH_2}; \\ \mathsf{CH_2CH=CH_2}, \ \mathsf{CH_2C=CH}, \ \mathsf{CH_2CO_2(alkyl)}, \ \mathsf{CH_2OCH_3}, \ \mathsf{CH_2CH_2}, \ \mathsf{CH_2CH_2CH_3}, \ \mathsf{CH_2CH_3}, \ \mathsf{CH_2CH_2CH_3}, \ \mathsf{CH_2CH_3}, \ \mathsf{CH_2CH_$

with the proviso that J is not

$$O \bigvee_{N-R}^{N-R}$$

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when:

A is oxygen and B is CR¹ double bonded to position 2; A is CH double bonded to position 2 and B is NR²; or A is NH and B is CR¹ double bonded to position 2.

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- 2. A compound of claim 1 in which A is nitrogen double-bonded to position 2 and B is oxygen.
- 3. A compound of claim 1 in which A is oxygen and B is CR¹ double bonded to position 2.
- 4. A compound of claim 1 in which A is NH and B is nitrogen double-bonded to position 2.
- 5. A compound of claim 1 in which A is nitrogen double bonded to position 2 and B is NR^2 .
- 10 6. A compound of claim 1 in which A is CH double bonded to position 2 and B is NR².
 - 7. A compound of claim 1 in which A is NH and B is CR¹ double bonded to position 2.
 - 8. A compound of claim 1 in which A and B are NH.
 - A compound having the formula

$$X_{(n)}$$
 Z

where X is selected from hydrogen, F, Cl, Br, alkyl, haloalkyl, CN, NO₂, and NH₂;

Y is selected from NO_2 , NH_2 , or -NHN=C(CH₃)R;

Z is selected from hydrogen, F, NH_2 , OH; with the proviso that when Y is -NHN=C(CH₃)R, Z is hydrogen;

n is 0-3;

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R is hydrogen, hydroxy, straight or branched chain lower alkyl, cycloalkyl, alkoxy, aryl, heteroaryl, alkenyl, haloalkyl, hydroxyalkyl, haloaryl, alkoxyaryl, arylalkyl, aryloxyalkyl, haloarylalkyl, alkylthio, heterocyclyl, alkoxyalkyl, arylcarbonyloxyalkyl, alkylcarbonyloxyalkyl, alkoxylalkyloxyalkyl, aminocarbonyloxyalkyl, aminoalkyl, cyanoalkyl, aminoalkenyl, carboxy, carboxyalkyl, alkylcarboxy, alkylcarboxyalkyl, formyl, aminocarbonyl, amino, oxygen, cyano, nitro, alkoxycarbonylamino, akylcarboxylalkoxy, alkylcarboxyoxyalkyl, alkvisulfonyl, (aryl)(alkoxy)alkyl, aryliminoalkyl, alkoxycarbonylalkylaminoalkyl, alkynylalkylthio, cyanoalkylthio, (aryl)(alkylcarbonyloxy)alkyl, arylalkoxyalkyl, alkoxycarbonylalkylthio, cyanothioalkyl, cyanothio, arylalkylthio, haloalkylalkynylalkylthio, alkenylalkylthio, aminocarbonylalkylthio, (hydroxy)(aryl)alkyl, arylalkylcarbonylaminoalkyl, aminocarbonyloxyalkyl, alkylcarbonylaminoalkyl, alkylsulfonylaminoalkyl, aminocarbonylalkyl, alkoxycarbonyl, and alkenyloxy, where the amino group may be substituted with one or two substituents independently selected from alkyl, hydroxy, alkoxy, carboxy, aryl, or alkylsufonyl;

J is selected from

 $\mbox{and} \quad \mbox{R}^{3} \ \mbox{is selected from hydrogen, alkyl, haloalkyl, $CH_{2}CN$,} \\ \mbox{CH}_{2}\mbox{CH}=\mbox{CH}_{2}, \mbox{CH}_{2}\mbox{C}\equiv\mbox{CH}, \mbox{CH}_{2}\mbox{CO}_{2}(\mbox{alkyl}), \mbox{CH}_{2}\mbox{OCH}_{3}, \mbox{and NH_{2}.} \\ \mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{CH}_{2}\mbox{C}=\mbox{C}=\mbox{C}=\mbox{C}=\mbox{C}=\mbox{C}=\mbox{C}=\mbox{C}=\mbox{C}=\mbox{C}=\mbox{C}=\mbox{C}=\mbox{C}=\mbox{C}=\mbox{C}=\mbox{C}=\mbox{C}=\mbox{C}=\mbox{C}=\mbox{C}$

- 10. An herbicidal composition comprising an herbicidally effective amount of a compound of claim 1, and an herbicidally compatible carrier therefor.
- amount of a compound of claim 1 and an herbicidally effective amount of one or more herbicides selected from the group consisting of glyphosate, 2,4-D, MCPA, MCPP, isoproturon, imazapyr, imazamethabenz, imazethapyr, imazaquin, acifluorfen, bifenox, fomasafen, ioxynil, bromoxynil, chlorimuron, chlorsulfuron, bensulfuron, pyrazosulfuron, thifensulfuron, triasulfuron, fenoxaprop, fluazifop, quizalofop, diclofop, bentazone, butachlor, metachlor, acetochlor, dimethenamide, dicamba, and fluroxypyr.
- 12. An herbicidal composition comprising an herbicidally effective amount of a compound of claim 1, and an herbicidally compatible carrier therefor.
- 13. A method of controlling undesired plant growth, comprising application to the locus where the undesired plants are growing or are expected to grow, an herbicidally effective amount of a composition of claim 1.
- 14. A method of controlling undesired plant growth, comprising application to the locus where the undesired plants are growing or are expected to grow, an herbicidally effective amount of a composition of claim 11.
 - 15. A compound having the formula

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where

A is oxygen and B is CR¹ double bonded to position 2; A is CH double bonded to position 2 and B is NR²; or A is NH and B is CR1 double bonded to position 2;

R is hydrogen, hydroxy, mercapto, straight or branched chain lower alkyl, cycloalkyl, alkoxy, aryl, heteroaryl, alkenyl, haloalkyl, hydroxyalkyl, haloaryl, alkoxyaryl, arylalkyl, aryloxyalkyl, haloarylalkyl, alkylthio, heterocyclyl, alkoxyalkyl, alkylcarbonyloxyalkyl, arylcarbonyloxyalkyl, alkoxylalkyloxyalkyl, 5 aminocarbonyloxyalkyl, aminoalkyl, cyanoalkyl, aminoalkenyl, carboxy, carboxyalkyl, alkylcarboxy, alkylcarboxyalkyl, formyl, aminocarbonyl, amino, oxygen, cyano, nitro, alkylsulfonylamino, alkylcarboxyoxyalkyl, aminosulfonyl, alkyisulfonyl, alkoxycarbonylalkylaminoalkyl, alkoxycarbonylamino, akylcarboxylalkoxy, (aryl)(alkylcarbonyloxy)alkyl, arylalkoxyalkyl, (aryl)(alkoxy)alkyl, aryliminoalkyl, 10 cyanothio, cyanothioalkyl, arylalkylthio, alkynylalkylthio, cyanoalkylthio, alkenylalkylthio, aminocarbonylalkylthio, alkoxycarbonylalkylthio, arylalkylcarbonylaminoalkyl, aminocarbonyloxyalkyl, haloalkylalkynylalkylthio, alkylcarbonylaminoalkyl. alkylsulfonylaminoalkyl, (hydroxy)(aryl)alkyl, aminocarbonylalkyl, alkoxycarbonyl, and alkenyloxy, where the amino group may be 15 substituted with one or two substituents independently selected from alkyl, hydroxy, alkoxy, carboxy, aryl, alkylsufonyl or haloalkylsulfonyl;

R¹ is hydrogen, lower alkyl, or haloalkyl;

R² is hydrogen, alkyl, haloalkyl, $CO_2(alkyl)$, $CH_2CO_2(alkyl)$, 20 $CH_2CONH(alkyl)$, $CH_2CON(alkyl)_2$, CH_2CO_2H , CH_2OCH_3 , $SO_2(alkyl)$, $CH_2CH=CH_2$, or $CH_2C=CH$;

X is selected from hydrogen, F, Cl, Br, alkyl, haloalkyl, CN, NO₂, and NH₂;

n is 0-3;

J is

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$$\begin{array}{c|c}
O & N & O \\
\hline
N-R^3 & \\
CF_2 & \end{array}$$

and R^3 is selected from hydrogen, alkyl, haloalkyl, CH_2CN , $CH_2CH=CH_2$, $CH_2C=CH$, $CH_2CO_2(alkyl)$, CH_2OCH_3 , and NH_2 .

In tional Application No PCT/US 98/03647

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IPC 6	FICATION OF SUBJECT MATTER C07D405/04 C07D403/04 C07D487/04 C07D513/04 A01N43/76 A01N43/66 o International Patent Classification (IPC) or to both it	A01N43/653	C07D251/ A01N43/5	16 C07D24	
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	ocumentation searched (classification system follow C07D A01N	ed by classification sym	bols)		
Documenta	tion searched other than minimumdocumentation to	the extent that such do	cuments are include	ed in the fields searc	hed
Electronic d	lata base consulted during the international search (name of data base and	, where practical, s	earch terms used)	
C. DOCUM	ENTS CONSIDERED TO BE RELEVANT				
Category °	Citation of document, with indication, where appro	priate, of the relevant p	passages		Relevant to claim No.
Y	EP 0 617 033 A (SUMITOMO September 1994 see claims	CHEMICAL CO)) 28		1-8, 10-15
Y	EP 0 561 319 A (SUMITOMO September 1993 see claims	CHEMICAL CO)) 22		1-8, 10-15
Y	EP 0 476 697 A (SUMITOMO March 1992 see claims	CHEMICAL CO)) 25		1-8, 10-15
Y	WO 95 05079 A (FMC CORP)	23 February	/ 1995		1-8, 10-15
	see claims	-/	-		
X Fur	ther documents are listed in the continuation of box	с. Х	Patent family m	embers are listed in	annex.
"A" docum consi "E" earlier filing "L" docum which citatic "O" docum other "P" docum	ategories of cited documents: nent defining the general state of the art which is not idered to be of particular relevance document but published on or after the international date ent which may throw doubts on priority claim(s) or is cited to establish the publicationdate of another on or other special reason (as specified) nent referring to an oral disclosure, use, exhibition or means the published prior to the international filing date but than the priority date claimed	"X" c "Y" c	or priority date and cited to understand invention document of particul cannot be consider involve an inventive document of particul cannot be consider document is combinents, such combin the art.	lar relevance; the cla	ne application but bry underlying the limed invention ne considered to limed invention set we step when the e other such docu- to a person skilled
	e actual completion of theinternational search			ne international searc	ch report
	12 June 1998		23/06/19	998 	
Name and	mailing address of the ISA European Patent Office, P.B. 5818 Patentlaan NL - 2280 HV Rijswijk Tel. (+31-70) 340-2040, Tx. 31 651 epo nl, Fax: (+31-70) 340-3016		Henry,	J	

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Int tional Application No
PCT/US 98/03647

Category °	ation) DOCUMENTS CONSIDERED TO BE RELEVANT Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Jategory *	Gradion of document, with indication, where appropriate, of the relevant passages	netevant to claim No.
Y	EP 0 373 461 A (NIHON TOKUSHU NOYAKU SEIZO KK) 20 June 1990 see claims	1-8, 10-15
Y	EP 0 304 920 A (NISSAN CHEMICAL IND LTD) 1 March 1989 see claims	1-8, 10-15
(EP 0 255 047 A (HOFFMANN LA ROCHE) 3 February 1988 see claims 21,22	9
х	DE 195 23 640 A (BAYER AG) 2 January 1997 see page 11; examples II-3	9
х	DE 195 04 188 A (BASF AG) 14 August 1996 see page 27	9
Ρ,Χ	WO 97 29105 A (KUMIAI CHEMICAL INDUSTRY CO) 14 August 1997 see abstract	1-8, 10-15
Р,Х	US 5 661 108 A (CRAWFORD SCOTT D ET AL) 26 August 1997 see the whole document	1-8, 10-15
Ρ,Χ	WO 97 42188 A (KUMIAI CHEMICAL INDUSTRY CO) 13 November 1997 see abstract	1-8, 10-15
Ρ,Χ	WO 97 08170 A (FMC CORP) 6 March 1997 see claims	1–15

Information on patent family members

In: tional Application No PCT/US 98/03647

Patent documer		Publication date	Patent family Publicatio member(s) date			
· · · · · · · · · · · · · · · · · · ·					06 02 1007	
EP 0617033	Α	28-09-1994	AU	676256 B	06-03-1997	
			AU	5784894 A	22-09-1994	
			BR	9401188 A	01-11-1994	
	-		CA	211 9047 A	18-09-1994	
	-		CN	1092774 A	28-09-1994	
		-	JP	6321941 A	22-11-1994	
			US	5411935 A	02-05-1995	
			US	5476834 A	19-12-1995	
			ZA	9401848 A	19-09-1994	
EP 0561319		22-09-1993	JP	5262765 A	12-10-1993	
			BR	9301204 A	21-09-1993	
			CA	2091505 A	19-09-1993	
			US	5308829 A	03-05-1994	
EP 0476697	A	25-03-1992	CA	2051942 A	22-03-1992	
			DE	69118070 D	25-04-1996	
			DE	69118070 T	21-11-1996	
			JP	5025165 A	02-02-1993	
			US	5169431 A	08-12-1992	
WO 9505079	- А	23-02-1995	AU	7480694 A	14-03-1995	
			US	5521147 A	28-05-1996	
EP 0373461	Α	20-06-1990	JP	2160781 A	20-06-1990	
EP 0304920	A	01-03-1989	CA	1309716 A	03-11-1992	
			CN	1032013 A,B	29-03-1989	
			ŲS	5039331 A	13-08-1991	
			JP	1250388 A	05-10-1989	
			JP	2666380 B	22-10-1997	
EP 0255047	 А	03-02-1988	AU	604250 B	13-12-1990	
			AU	7637187 A	11-02-1988	
			CA	1286662 A	23-07-1991	
			DK	366887 A	13-05-1988	
			ÜS	4859229 A	22-08-1989	
			JP	63041466 A	22-02-1988	
			ZA	8705466 A	02-02-1988	

Information on patent family members

In itional Application No
PCT/US 98/03647

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
DE 19523640	A	02-01-1997	AU 6304396 A WO 9701542 A EP 0835247 A	30-01-1997 16-01-1997 15-04-1998
DE 19504188	Á-	14-08-1996	AU 4621596 A CA 2210909 A WO 9624590 A EP 0808310 A	27-08-1996 15-08-1996 15-08-1996 26-11-1997
WO 9729105	Α	14-08-1997	AU 1670297 A	28-08-1997
US 5661108	Α	26-08-1997	NONE	
WO 9742188	Α	13-11-1997	AU 2651297 A JP 10053584 A	26-11-1997 24-02-1998
WO 9708170	Α	06-03-1997	AU 6961896 A	19-03-1997